

EDC  WFD



**Training workshop: Solutions to tackle WFD  
requirements for estrogen determination in water**

7-9 September 2022

EDC  WFD

BEFORE TO START

THIS TRAINING WILL BE REGISTERED

DOES ANYONE HAVE AN OBJECTION?

- This **Training/Workshop** aims:
  - to present the **knowledge gained from the EDC-WFD** project whose objective is to develop reliable and harmonized measurement methods for estrogens, which are key Endocrine Disrupting Chemicals (EDC), to comply with Water Framework Directive requirements
  - to **accelerate the transfer of the most promising measurement methods and methodologies to interested parties**: laboratories, PT providers, researchers
- The training workshop will cover **all aspects of measurements** from sampling to final method validation and will address both Mass spectrometry based methods as well as incoming Effect Based Methods (in vitro bioassays)

## 7th of September Session 1

**09:00 - 09:10: Welcome address**

**09:10 - 09:50: Presentation of the project and context**

**09:50 - 10:20: Issues and challenges related to estrogen analysis in relation to the WFD**

**10:20 - 11h00: Challenges related to sampling**

***11:00 - 11:15: Break***

**11:15 - 11:35: Overview of quantification strategy**

**11:35 - 12:15: Sample preparation**

## **8th of September Session 2**

**09:00 - 09:30: Discussion forum / debriefing from day 1**

**09:30 - 10:30: Mass spectrometry methods - Instrumental  
developments**

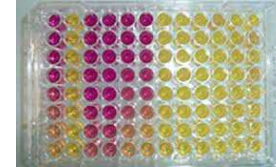
***10:30 - 10:45: Break***

**10:45 - 11:45: Achievements of Mass spectrometry based methods  
\_ method performances and measurement reliability**

**11:45 - 12:00: Concluding remarks**

**12:00 - 12:15: Next step \_ Towards Interlaboratory Comparison**

**9th of september**  
**Session 3 dedicated to Effect Based Methods (EBM)**



**09:00 - 09:10 : Welcome address**

**09:10 - 09:40 : Presentation of the project and context**

**09:40 - 10:05 : Context and presentation of EBM methods versus MS based methods**

**10:05 - 10:40 : EBM protocols**

**10:40 - 11:15 : EBM data treatments**

***11:15 - 11:30 : Break***

**11:30 - 11:45 : Concluding remarks**

**11:45 - 12:00 : Next step : Towards Interlaboratory Comparison**



EURAMET is...



- The gateway to Europe's integrated metrology infrastructure
- It facilitates access to European measurement expertise, and underpins the delivery of globally competitive, high impact metrology
- It ensures Europe maintains its global reputation for excellence in measurement science
- It raises awareness of the value of measurement by demonstrating the impact of metrology on society's grand challenges



# European Association of National Metrology Institutes



## Members:

38 European NMIs  
28 of them are participating in EMPIR

## Associates:

78 DIs (Designated Institutes)

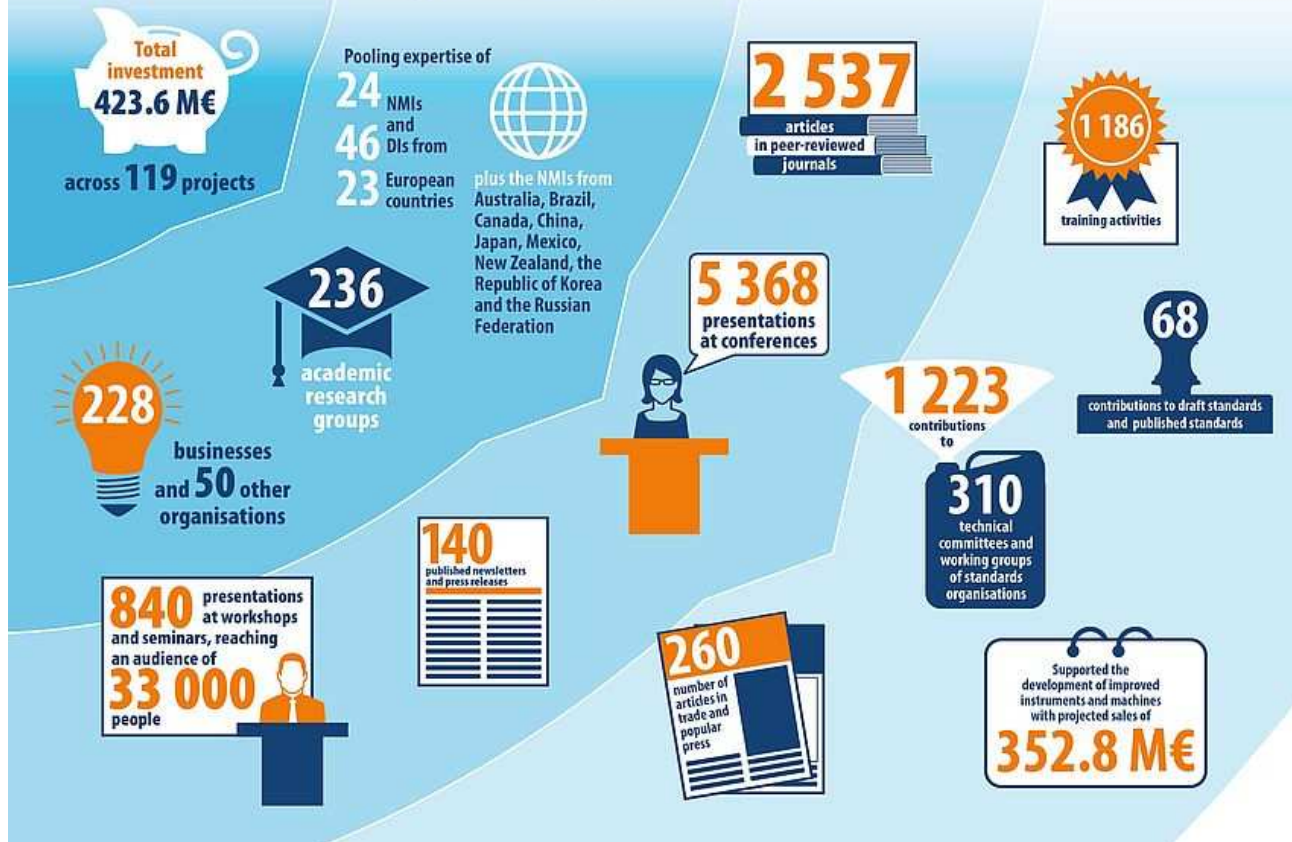
## metrologists:

NMIs: ~ 5000  
DIs: ~ 1500



<https://www.euramet.org/about-euramet/>

## EURAMET's European Metrology Research Programme at a glance



<https://www.euramet.org/metrology-for-societys-challenges/>

## What is EMPIR?



- It is about improving measurement to drive innovation and competitiveness.
- It enables European metrology institutes, industrial organisations and academia to collaborate on joint research projects.
- It is implemented by EURAMET (European Association of National Metrology Institutes).
- It is based on Article 185 of the Lisbon Treaty.
- It is jointly funded by the EMPIR participating countries and the European Union and has a budget of approximately 600 M€ over seven years.

<https://www.euramet.org/research-innovation/research-empir/about-empir/>

## Objectives for EMPIR



**Challenge /problem:** increasing demands on metrology infrastructure alongside static or reducing national budgets. This requires strategically planned metrology R&D (reducing fragmentation and duplication to achieve critical mass) to ensure delivery of future metrology capabilities / services to meet European needs

### EMPIR

#### High-level objectives

European growth and jobs  
Respond to societal challenges  
Create an integrated European Research Area

#### Specific objectives

- Boost industrial uptake of metrology research supporting development of new & improved products and services
- Improve standardisation / regulation
- Underpin a coherent, sustainable and integrated European metrology landscape

#### Operational objectives

- Develop common research agenda
- Support European collaboration
- Fund projects that support innovation, regulation, societal challenges
- Capacity building across Europe
- Efficient & effective programme management

## EMPIR objectives



### **Boost industrial uptake and improve standardisation**

- At least 400 M€ of European turnover from new or significantly improved products and services that can be attributed to the research activities of EMPIR and its predecessors.

**346 M€ identified so far.**

- At least 60% of CEN/CENELEC /ISO/IEC Technical Committees and equivalent standardisation bodies with potential to benefit directly from EMPIR projects to engage with the programme.

**Contributions to 17 published and 103 draft standards.**

## EMPIR objectives



### Underpin a coherent, sustainable and integrated European metrology landscape to fully exploit the EU potential

- Maintain a level of at least 50 % of dedicated national metrology research investments in Europe being coordinated or influenced via the programme. **Level maintained**
- All European NMIs and their designated institutes to interact with the programme. **Cyprus, Malta**
- European leadership in at least 20 % of international metrology committees. **50 %**

## EUROPEAN PARTNERSHIP ON METROLOGY



- ❑ The EPM will bring together the measurement science community and stakeholders to deliver on global challenges including health and climate, support the European Green Deal, and underpin innovation in industry through collaborative research.
- ❑ The EPM aims to **support accelerating the transition towards a green, climate neutral and digital Europe**, as well as strengthening the resilience, competitiveness, and economic growth of the European industry.
- ❑ European Partnerships are a key implementation tool of the European Commission's Horizon Europe; an ambitious research and innovation programme, running from 2021 to 2027.

## EUROPEAN PARTNERSHIP ON METROLOGY



- ❑ The EPM is co-funded by the Member States and the European Union with an expected budget of over **650 million euro**. Its expected impact is manifold, as it will support a wide range of policies, commerce and advancement of key European challenges.
- ❑ It builds on the progress achieved under the previous European Metrology Research Programme, and aims to break new ground by contributing to the development of self-sustaining, coordinated metrology infrastructures, with the capacity to continue joint research and innovation after 2030.

<https://www.euramet.org/research-innovation/metrology-partnership/>



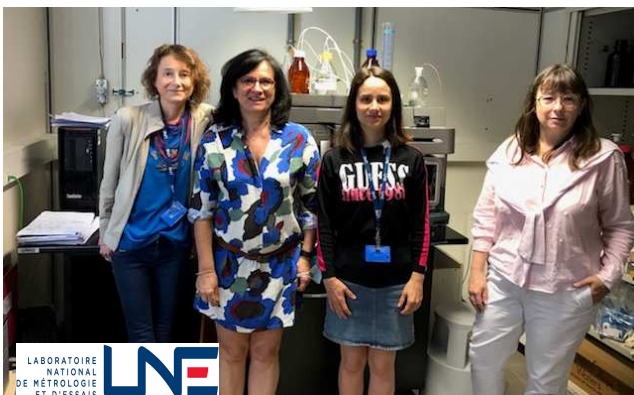
# OVERVIEW OF THE PROJECT

## THE CONSORTIUM

- 8 partners / 6 European countries
- 5 NMI/DI, one DI operating outside of its designation, one academic research laboratory, one research institute
- Consortium brings together **scientific excellence** in research institutes and experience in **ultra-trace measurements of micropollutants**
- **Balance of expertise:** development and certification of RM, proficiency tests / interlaboratory comparison design, method development and validation, standardisation



Start date: 1<sup>st</sup> September 2019  
Duration: 36+6 months  
Budget: 800K€



S. LARDY-FONTAN, V. LE DIOURON, C. FALLOT,  
B. LALERE & our LC/MS<sup>2</sup> (critical member of the team)



K. LE MENACH, P. PARDON, H. BUDZINSKI



C. PIECHOTTA, L. STEINHAÜSER, T. WESTPHALEN,  
U.-A. KLYK-SEITZ, K. KAMINSKI, S. KLUGE



T. NÄYKKI, J. VIIDANOJA



E. HEATH, A. KOVACIC



T. GÖKCEN, I. UN



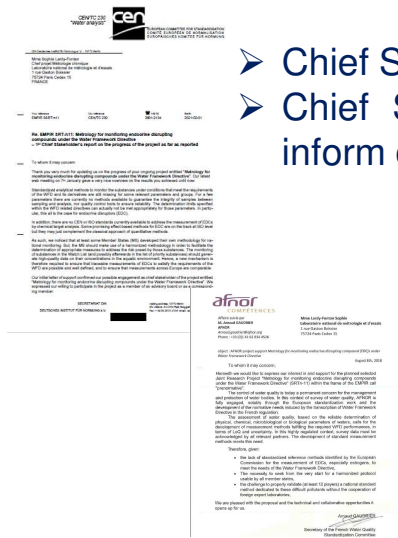
P. De ZORZI, M. POTALIVO, C. ASCENZI, E. CALABRETTA, S. BARBIZZI, G. MOLTEDO



# THE PROJECT

## AIMS:

- ❖ Address the standardization lack for harmonised measurement methods for estrogens in whole water
- ❖ Ensure that measurements of estrogens are traceable, well defined, meet the requirements of the WFD, and thus are comparable across Europe (and worldwide)



- Chief Stakeholder: Ulrich BORCHERS Chairman of CEN/TC 230
- Chief Stakeholder and DIN secretary (Andreas Paetz) are kept inform of the progress

- Support from AFNOR (Arnaud Gaudrier: Secretary of SC «Water Quality»)

## CONTEXT

- **Estrogens:** group of chemicals of similar structure mainly responsible for female sexual development and reproduction.
- In water ecosystem:
  - **Pseudo-ubiquitous** and occur at **ultra-trace level** ( $\ll \text{ng L}^{-1}$  to tens  $\text{ng L}^{-1}$ )
  - Level at which they can have effects in natural species
    - **Threat for biodiversity**



- **No EN or ISO standard for MS-based methods** currently available or in progress
- (Accredited) testing laboratories **develop and validate in-house methods** according to internal criteria
- Most of (accredited) testing laboratories **failed to achieve the very low LOQ** to enable monitoring of estrogens at relevant level
- **Metrological endpoints** have been highlighted of particular importance if effect-based method (EBM) results are to be used in a regulatory context
- **Lack/absence of reference materials and ad'hoc proficiency tests**

## AIM

Natural and pharmaceutical estrogens are key Endocrine Disrupting Chemicals (EDC) which are monitored differently depending on the country, and for which standardised reference methods are currently not available.

⇒ Main Objective: Develop reliable and harmonized measurement methods for estrogens, to comply with the WFD Directive requirements

⇒ **Outcomes: to be disseminated to CEN/ TC 230 and ISO/ TC 147 to be fed into the documentary standards they develop**

## OBJECTIVES

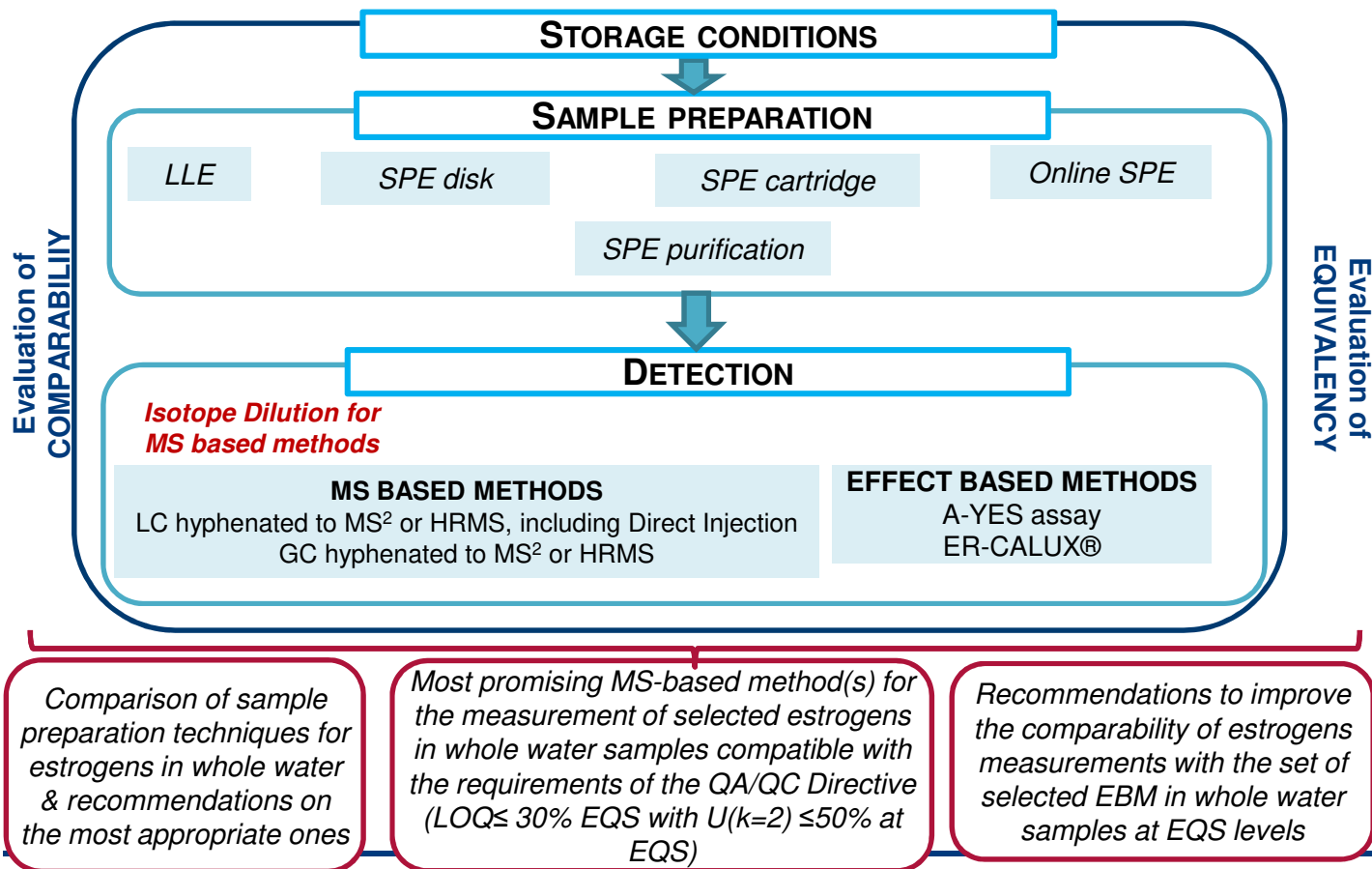
- The overall objective of this project is to **develop traceable measurement methods** for endocrine disrupting chemicals, with a specific focus **on three estrogens of the first watch list: 17-beta-estradiol (17 $\beta$ E2), 17 alpha ethinylestradiol (17 $\alpha$ EE2), and estrone (E1))**
- Estrogens 17-alpha-estradiol (17 $\alpha$ E2) and estriol (E3) will be included to demonstrate the **reliability of the developed methods**
  - ⇒ **to support the requirements of Directive 2013/39/EC, Directive 2009/90/EC and Commission Implementation Decision (EU) 2018/840,**
  - ⇒ **improve the comparability and compatibility of measurement results within Europe**



■ The specific objectives of the project are to:

- 1. Optimize and validate traceable aqueous reference Mass Spectrometry-based methods for the analysis of 5 estrogenic compounds** prioritizing  $17\beta\text{E}2$ ,  $17\alpha\text{EE}2$ , and E1 in **whole water samples** at environmental quality standard (EQS) levels. **Methods will have limit of quantification (LOQ) not exceeding 30% EQS with a measurement uncertainty of  $\leq 50\%$  at EQS**
- 2. Develop production methods for aqueous reference materials (RM)**, which are as close as possible to real water samples, with proven homogeneity, short- and long-term stability
- 3. Improve the comparability of estrogen measurements with selected Effect-Based Methods (EBM)** in whole water samples at EQS level. Methods will have been correctly calibrated and information on uncertainty will be provided
- 4. Organize and perform an interlaboratory comparison (ILC)** to demonstrate the performance of the developed methods using the reference material (RM) for the selected estrogen substances
- 5. Contribute to the work of key European and international standardization organizations e.g. CEN TC 230 and ISO TC 147**

## METHOD OPTIMIZATION





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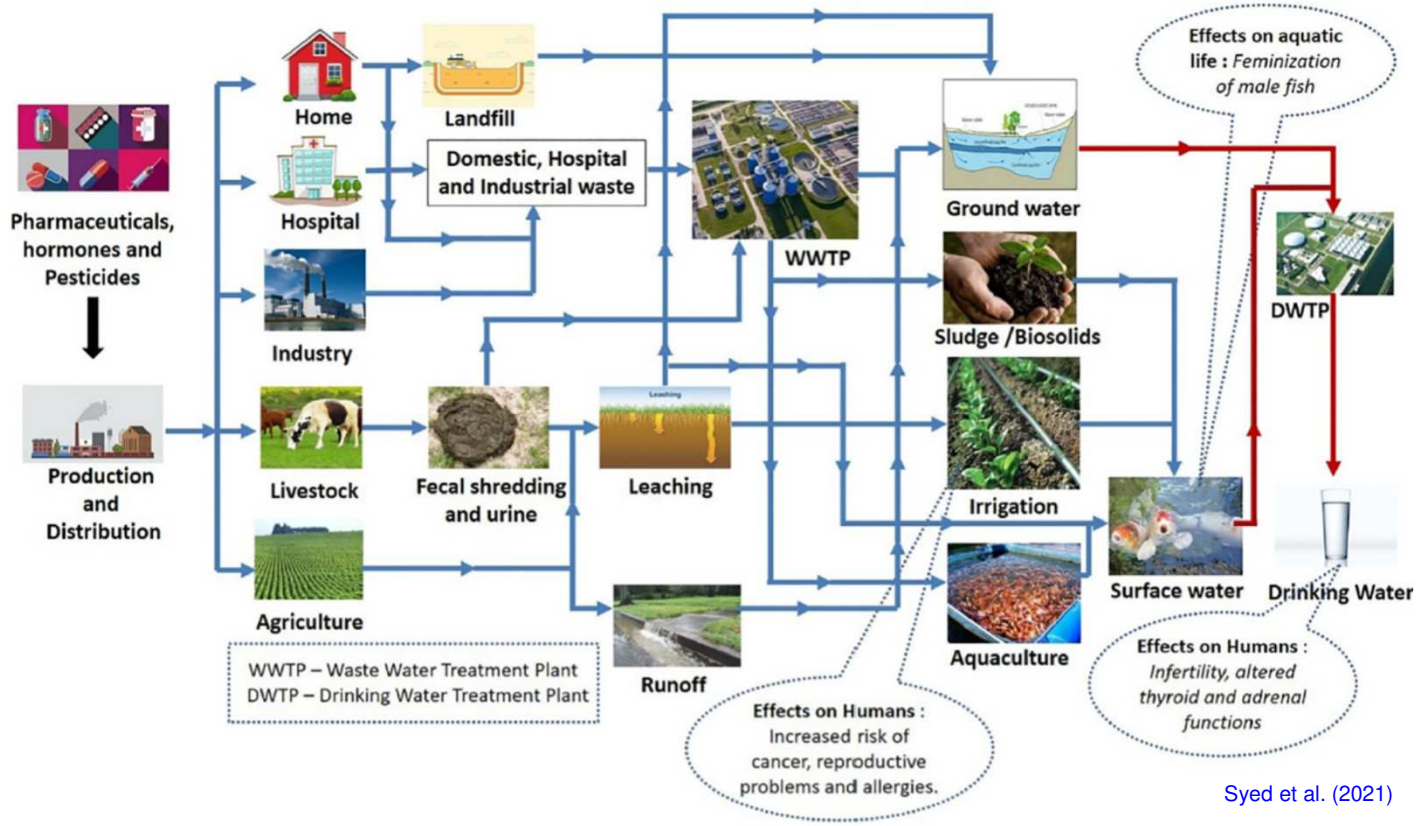
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7-9 september 2022

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# GENERALITIES

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE



To learn more (Review Aris et al., 2014; Adeel et al., 2017)

## BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

	17βE2	E1	17αEE2	E3
Water solubility (mg L <sup>-1</sup> )	3.6	760 (20°C) 30 (25°C)	11.3 at 27°C 4.7 at 20°C 19 at 20°C	27.34 at 25 °C
Volatilisation	According to vapor pressure and Henry constant values, the substance is not likely to volatilise from water phase.			
Organic carbon – water partition coefficient (K <sub>OC</sub> )	K <sub>OC</sub> = 791.7 log K <sub>OC</sub> = 2.9	log K <sub>OC</sub> = 3.019 log K <sub>OC</sub> = 4.375	Log K <sub>OC</sub> = 2.92 – 4.68 K <sub>OC</sub> = 192 – 2 955 log K <sub>OC</sub> = 3.21 – 5.44 K <sub>OC</sub> = 1 622 – 275 423	K <sub>OC</sub> = 1200
Sediment – water partition coefficient (K <sub>susp-water</sub> )	251		25 – 34 429	log K <sub>d</sub> = 1.33
Octanol-water partition coefficient (Log K <sub>ow</sub> )	4.01	3.13 3.43	3.67 4.2, 25°C, pH 7	2.45

- Estrogens typically have both low water solubility and mid K<sub>ow</sub> ⇒ hydrophobic + high potential to bind to sediments/ SPM
- Photolysis, biodegradation, and sorption are the likely leading attenuation pathways controlling the fate of estrogens in water ecosystem

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

## Sorption

- Yu et al. (2004) and Brett et al. (2014) observed that in one day, 80-90% of E1 and EE2 can be sorbed to suspended solids and reach steady-state within ten days
- Estrogens can also sorb to humic acids (HA) in water (Chowdhury et al., 2011)
- The presence of saline compounds such as NaCl induces flocculation and aggregation, which can promote sorption processes (Lai et al., 2000; De Mes et al., 2005).
- Estrogen sorption capacity is also positively related to total organic carbon (TOC) content since the sorption occurs through hydrogen bonding reactions between organic carbon and estrogen compounds (Lai et al., 2000; Nghiem et al., 2004; D'Alessio et al., 2014).
- Yu et al. (2004) reported a strong sorption competition effect between one estrogen compound and other estrogens or hydrophobic compounds for sorption sites. The competition is most significant when the concentration of the primary sorbate is low and the concentration of the competitive sorbate is high (Lai et al., 2000; Yu et al., 2004)



## BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

### Sorption

- The isomer-specific sorption of E2 isomers is dominated by H-bonding and aromatic interaction which  $17\beta$ -E2 sorption was preferential than  $17\alpha$ -E2 (Qiao et al. 2011). The orientation of the –OH group at the C17 position in the D-ring plane may favor the sorption of  $17\beta$ -E2 while that of  $17\alpha$ -E2 is outside the plane.
- The sorption process of estrogens is associated with organic compounds, colloids, composition of clay minerals, specific surface area (SSA), cation exchange capacity (CEC), and pH value in the environment (Yu et al., 2019)

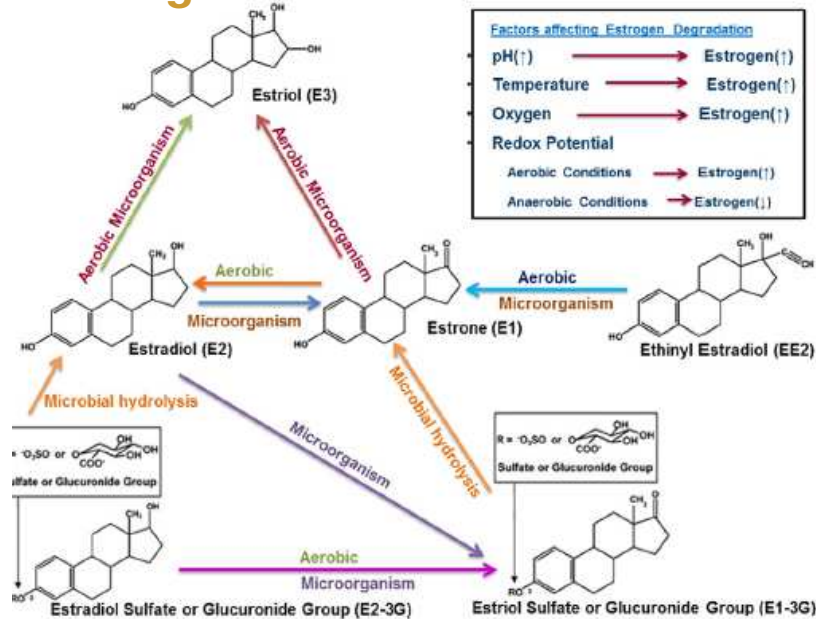
## BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

### Photodegradation

- Estrogens such as E2 and EE2 in aquatic environments are also susceptible to breakdown by photocatalysis and photolysis. The extent of degradation by photolysis and photocatalysis depends upon an estrogen's chemical structure.
- The photolysis process of estrogens is affected by humic substances under simulated sunlight
- [Zuo et al., 2013](#): EE2 degradation was studied in a lake site in the USA. Under aerobic conditions, half-life was estimated as 108 days. However, under natural sunlight, photo degradation accounted for a much shorter half-life of 23 h

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

## Biodegradation



- Estrone (E1), estradiol (E2), and estriol (E3) lie on interconnecting metabolic pathways
- Microorganisms living in aerobic and anaerobic conditions can convert one estrogen to another
- Even EE2 can be converted to E1

Table 5  
Half-life (in days) of steroidal estrogens from different aquatic sources.

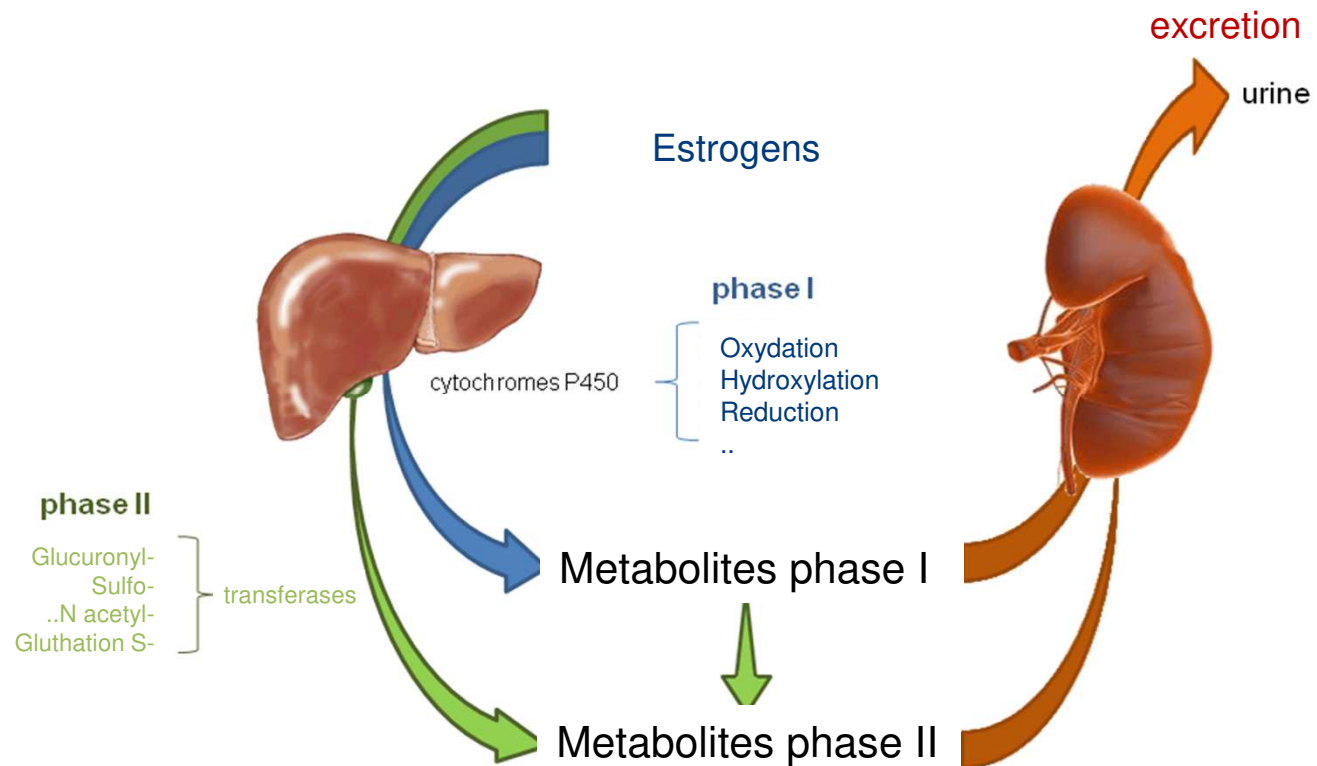
Medium	Oxygen status	E1	17β-E2	E3	EE2	Reference
Natural soil	Aerobic	2.8–4.9	0.8–1.1	0.7–1.7	NDA	Biswas et al., 2013
Aire river water	Aerobic	2.4	2.3	NDA	NDA	Jürgens et al., 2002
Calder river water	Aerobic	2.2	2.7	NDA	NDA	
Thames river water	Aerobic	3.1	4.1	NDA	NDA	
River water	NDA	2–3	2–3	NDA	4–6	Ying et al., 2002
Sandy-loam	Anaerobic	NDA	NDA	NDA	NDA	Ying and Kookana, 2005
Aquifer materials	Aerobic	NDA	NDA	NDA	81	Ying et al., 2003
Natural water	Aerobic	NDA	NDA	NDA	1.5	Zheng et al., 2011

Adeel 2017

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

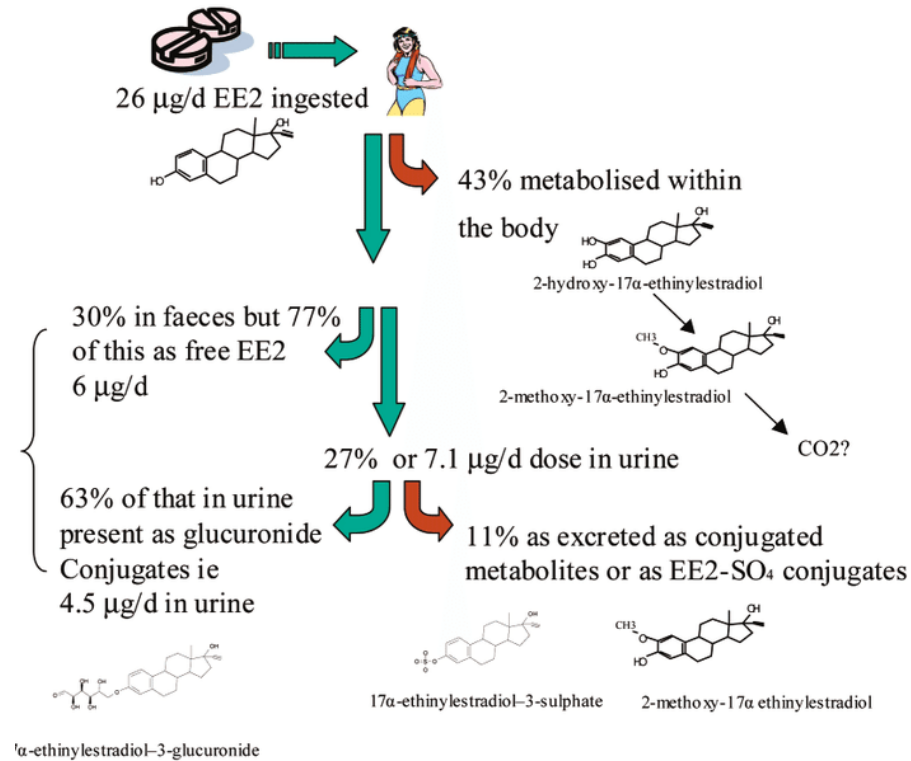
## Metabolisation

Excretion by humans and animals in complex mixtures of free and conjugated forms



# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

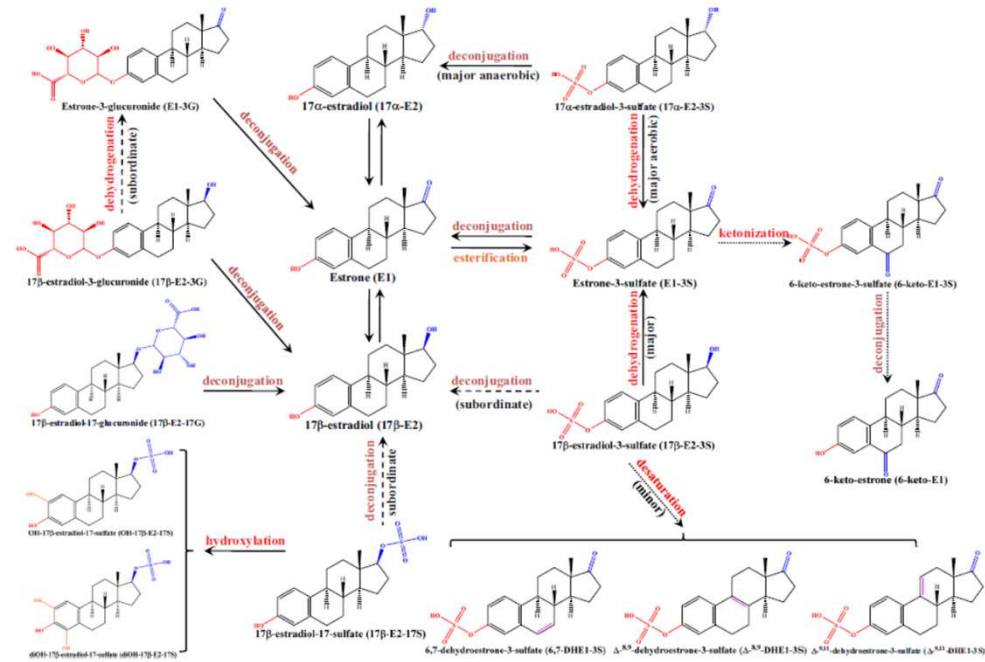
## Metabolisation



Johnson et Richard (2004)

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

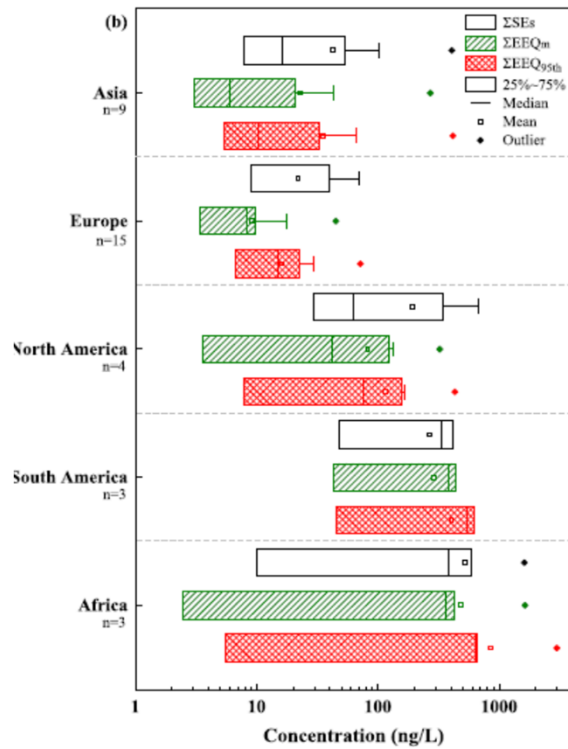
## Metabolisation



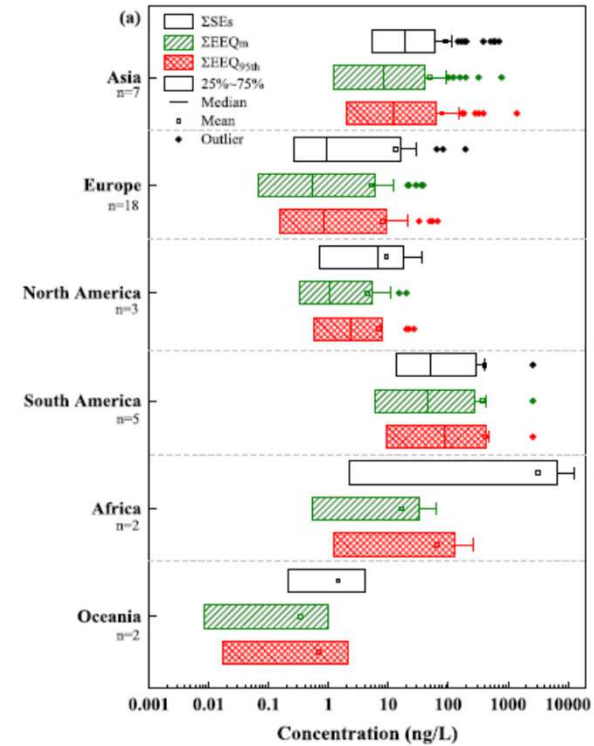
- Chronic release to the environment
- Conjugation is reversible and could lead to the formation of free conjugated forms or new transformation products (eg [Ma et al 2018](#), [Yu 2019](#) )

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

Wastewater treatment plants effluent



Natural water body

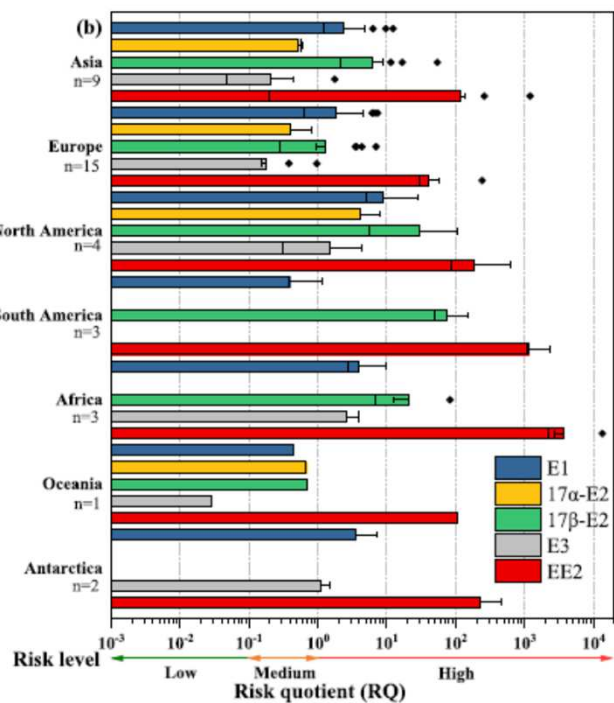


⇒ **Worldwide contamination**

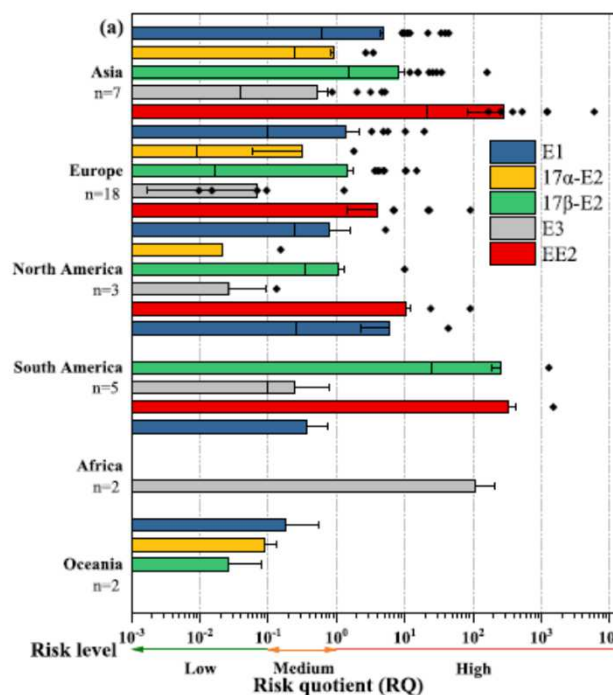
Du et al. 2020

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

Wastewater treatment plants effluent



Natural water body



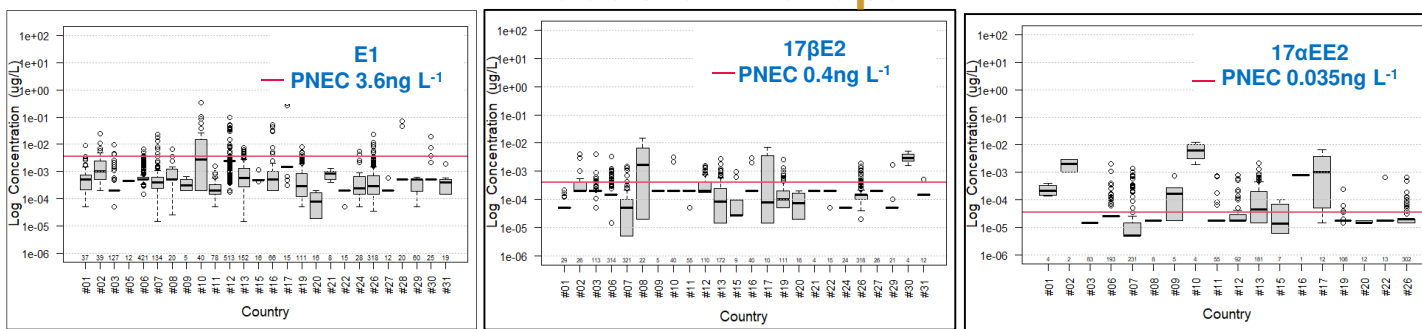
⇒ Water bodies at risks

Du et al. 2020



# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

## Focus on Europa



Statistics for MEC according to different choices (<LOQ replaced by 50% LOQ as setted by Directive QA/QC)

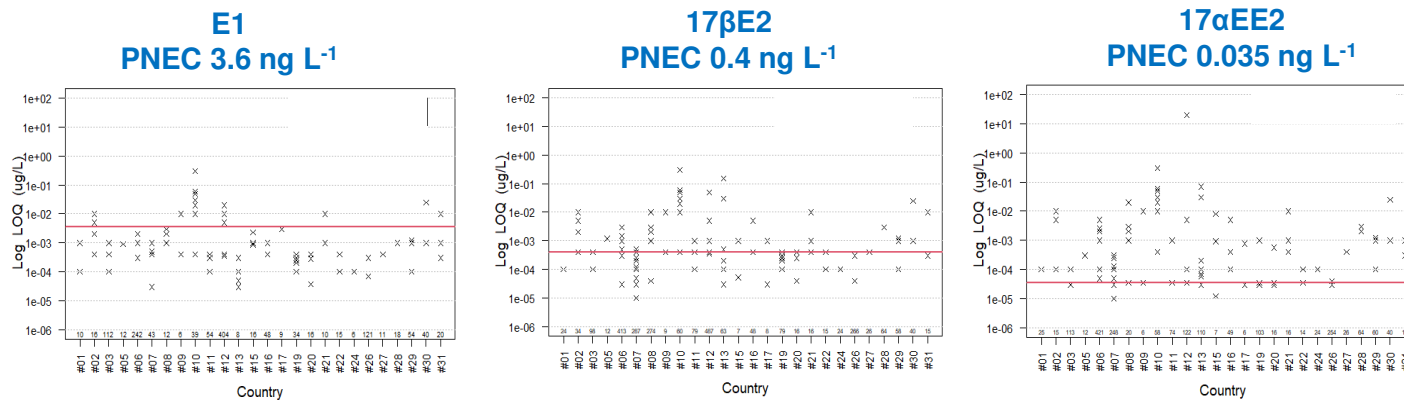
Concentration (µg L <sup>-1</sup> )	E1	17βE2	17αEE2
Min	1.50E-05	5.00E-06	0,000005
Mean	2.39E-03	2.54E-04	9.81E-05
StDev	1.61E-02	7.53E-04	5.51E-04
Median	5.00E-04	1.50E-04	1.75E-05
P90	0.0025	3.20E-04	0.00013
P95	0.005	7.92E-04	0.0003
P99	0.03289	3.08E-03	0.00139
Max	0.481	0.015	0.0125

EQS suggestion EE2--JRC-2021-DRAFT; EQS E2 & E1-JRC-2021-FINAL DRAFT

# BIOGEOCHEMICAL CYCLE: FATE AND OCCURRENCE

## Focus on Europa

Range of LOQs for non-quantified samples in combined dataset per country.



- EU-wide comprehensive assessment of the WL compounds E1, E2 and especially EE2 is not feasible in the current situation

EQS suggestion EE2--JRC-2021-DRAFT; EQS E2 & E1-JRC-2021-FINAL DRAFT

# REGULATORY CONTEXT

## REGULATORY CONTEXT

- **Directive 2000/60/EC** of the European Parliament and of the Council establishing a framework for the Community action in the field of water policy“  
⇒ precise timetable, with 2015 as the deadline for getting all European waters into good condition
- **Directive 2008/105/EC** of the European Parliament and of the Council of 16 December 2008 on environmental quality standards in the field of water policy ⇒ lays down environmental quality standards (EQS) for priority substances and certain other pollutants
- **Directive 2009/90/EC “QA/QC”** ⇒ lays down technical specifications for chemical analysis and monitoring of water status
- **Directive 2013/39/EU** amending the Environmental Quality Standards Directive 2008/105/EC ⇒ introduces the “Watch List” monitoring mechanism in order to collect high-quality Union-wide monitoring data for the purpose of supporting future prioritisation exercises.

## REGULATORY CONTEXT

- **COMMISSION IMPLEMENTING DECISION (EU) 2015/495** of 20 March 2015 ⇒ establishes a watch list of substances for Union-wide monitoring in the field of water policy

ANNEX

Watch list of substances for Union-wide monitoring as set out in Article 8b of Directive 2008/105/EC

Name of substance/group of substances	CAS number <sup>(1)</sup>	EU number <sup>(2)</sup>	Indicative analytical method <sup>(1)</sup> <sup>(4)</sup> <sup>(5)</sup>	Maximum acceptable method detection limit (ng/l)
17-Alpha-ethinylestradiol (EE2)	57-63-6	200-342-2	Large-volume SPE — LC-MS-MS	0,035
17-Beta-estradiol (E2), Estrone (E1)	50-28-2, 53-16-7	200-023-8	SPE — LC-MS-MS	0,4

- **COMMISSION IMPLEMENTING DECISION (EU) 2018/840** of 5 June 2018 establishing a watch list of substances for Union-wide monitoring in the field of water policy pursuant to Directive 2008/105/EC of the European Parliament and of the Council and repealing Commission Implementing Decision (EU) 2015/495 ⇒ Maintain in the list because of insufficient data quality

## REGULATORY CONTEXT

➤ **DECISION TO UPDATE THE LISTS OF PRIORITY ENDORSED BY COMMISSION** ⇒ Shall be finalized by end of 2022

Inland waters	Method Detection Limit Watch List ( $\mu\text{g L}^{-1}$ ) $\approx$ PNEC	EQS December 2021 scientific opinion by SCHEER ( $\mu\text{g L}^{-1}$ )*
17 $\alpha$ EE2	0.000035	0.000017
17 $\beta$ -E2	0.0004	0.00018
E1	3.6	0,00036

Salted waters	Method Detection Limit Watch List ( $\mu\text{g L}^{-1}$ ) $\approx$ PNEC	EQS December 2021 scientific opinion by SCHEER ( $\mu\text{g L}^{-1}$ )*
17 $\alpha$ EE2	No specification	0.0000016
17 $\beta$ -E2		0.000009
E1		0,000018

Scientific Opinion on "Draft Environmental Quality Standards for Priority Substances under the WFD"-17-Alpha-Ethinylestradiol (EE2), Beta-Estradiol (E2) and Estrone (E1) [https://ec.europa.eu/health/publications/scientific-opinion-draft-environmental-quality-standards-priority-substances-under-wfd-17-alpha\\_en](https://ec.europa.eu/health/publications/scientific-opinion-draft-environmental-quality-standards-priority-substances-under-wfd-17-alpha_en)

## REGULATORY CONTEXT

- **DECISION TO UPDATE THE LISTS OF PRIORITY ENDORSED BY COMMISSION** ⇒ Shall be finalized by end of 2022
- EU-wide comprehensive assessment of the WL compounds E1, E2 and especially EE2 is not feasible in the current situation.
- However, when employing *in vitro* EBMs together with respective substance specific Effect-Based Trigger (EBT) values, it is possible to discriminate between a sufficient and insufficient chemical status, as defined by the suggested EQSs for E1, E2 and EE2, with sensitivities and specificities near 90%.
- Moreover, EDC-WFD project has been launched in 2018. Within the project four *in vitro* EBMs will be validated according to QA/QC Directive requirements and the comparability of estrogen measurements with the selected EBMs in whole water samples at EQS level will be assessed and improved. The assessment of metrological references will demonstrate their accuracy sustaining the *In vitro* EBMs suitability for the detection of estrogenicity for trend monitoring, status assessments, prioritisation of water bodies, identification of sources and investigative monitoring.

## REGULATORY CONTEXT

- **DECISION TO UPDATE THE LISTS OF PRIORITY ENDORSED BY COMMISSION** ⇒ Shall be finalized by end of 2022
- *In vitro* EBMs for the detection of the ER activation cover a relevant Mode of Action.
- SOPs for this type of bioassays are available and three assays are even ISO standardised. Further validation and interlaboratory studies for other bioassays evaluating effects of estrogenic compounds would provide a wider choice of methods.
- **Annex VIII** of the WFD was developed to also target the substances that these EBMs respond to. In the short term (2019 WFD review), it could be considered to rephrase the Annex **allowing the use of *in vitro* EBMs to assess the presence of substances causing effects on endocrine-related functions.**
- However, field studies should be performed to evaluate the potential of these EBMs to identify sources of emission as a basis for subsequent measures for improvement. *In vitro* EBMs evaluated with suitable EBT values can screen, with a high level of safety, water samples for further chemical analyses. **Medium-term (next mandate) *in vitro* EBMs for the detection of ER activation might be included in future WL program after the development of guidance documents and a comparison of suitable bioassays.**



# DIRECTIVE QA/QC

EN

Official Journal of the European Union

L 188/11  
1.8.2009

COMMISSION DIRECTIVE 2009/90/EC  
of 31 July 2009

laying down, pursuant to Directive 2000/60/EC of the European Parliament and of the Council,  
technical specifications for chemical analysis and monitoring of water status  
(Text with EEA relevance)

➤ Establishes **minimum performance criteria for methods of analysis** to be applied by Member States when monitoring water status, sediment and biota, as well as **rules for demonstrating the quality of analytical results**.

➤ Article 2 definitions

- **limit of quantification**' means a stated multiple of the limit of detection at a concentration of the determinand that can reasonably be determined with an acceptable level of **accuracy (trueness) and precision**. The limit of quantification can be calculated using an appropriate standard or sample, and may be obtained from the lowest calibration point on the calibration curve, excluding the blank
- **uncertainty of measurement**' means a non-negative parameter characterizing the dispersion of the quantity values being attributed to a measurand, based on the information used

# DIRECTIVE QA/QC

EN

Official Journal of the European Union

L 8.2009

COMMISSION DIRECTIVE 2009/90/EC  
of 31 July 2009

laying down, pursuant to Directive 2000/60/EC of the European Parliament and of the Council,  
technical specifications for chemical analysis and monitoring of water status

(Text with EEA relevance)

- Article 3 Methods of analysis
  - All methods of analysis, are validated and documented in accordance with EN ISO/IEC-17025 standard or other equivalent standards accepted at international level.
- Article 4 Minimum performance criteria for methods of analysis
  - the minimum performance criteria for all methods of analysis applied are based on
    - | an uncertainty of measurement of 50 % or below ( $k= 2$ ) estimated at the level of relevant EQS and
    - | a limit of quantification equal or below a value of 30 % of the relevant EQS
  - In the absence of relevant EQS for a given parameter, or in the absence of method of analysis meeting the minimum performance criteria monitoring is carried out using best available techniques not entailing excessive costs.

## DIRECTIVE QA/QC

EN

Official Journal of the European Union

L 188/1

COMMISSION DIRECTIVE 2009/90/EC  
of 31 July 2009

laying down, pursuant to Directive 2000/60/EC of the European Parliament and of the Council,  
technical specifications for chemical analysis and monitoring of water status  
(Text with EEA relevance)

- Article 6 Quality assurance and control
  - laboratories or parties contracted by laboratories apply quality management system practices in accordance with **EN ISO/IEC-17025 or other equivalent standards** accepted at international level
  
  - laboratories or parties contracted by laboratories demonstrate their competences in analysing relevant physico-chemical or chemical measurands by:
    - | participation in **proficiency testing programmes** at levels of **concentrations that are representative** of chemical monitoring programmes
    - | analysis of available **reference materials** that are representative of collected samples + appropriate levels of concentrations

| No requirements on traceability to SI

I

(Legislative acts)

DIRECTIVES

DIRECTIVE (EU) 2020/2184 OF THE EUROPEAN PARLIAMENT AND OF THE COUNCIL  
of 16 December 2020  
on the quality of water intended for human consumption  
(recast)  
(Text with EEA relevance)

Brussels, 19.1.2022  
C(2022) 142 final



COMMISSION IMPLEMENTING DECISION

of 19.1.2022

establishing a watch list of substances and compounds of concern for water intended for human consumption as provided for in Directive (EU) 2020/2184 of the European Parliament and of the Council



ANNEX

WATCH LIST OF SUBSTANCES AND COMPOUNDS OF CONCERN FOR WATER INTENDED FOR HUMAN CONSUMPTION

Name of substance/group of substances or compound/group of compounds	CAS number	EU number	Guidance value (ng/L)	Limit of Quantification <sup>1</sup> (ng/L)	Possible method of analysis
17-beta-estradiol	50-28-2	200-023-8	1	≤ 1	-
nonylphenol <sup>2</sup>	84852-15-3	284-325-5	300	≤ 300	EN ISO 18857-2



# WHAT DO WE NEED TO MEASURE? WHAT ARE WE MEASURING?

## WHOLE WATER MEASUREMENTS WHAT ARE WE TALKING ABOUT ?

### Definitions

- **Whole water** is synonym for the original water sample and shall mean the water sample when solid matter and the liquid phase have not been separated [Guidance CIRCA 19]
- **Liquid (dissolved) fraction** shall mean an operationally defined fraction of whole water from which suspended particulate matter has been removed by an appropriate methodology. [Guidance CIRCA 19]
- **Suspended particulate matter** shall mean the particulate matter fraction of the whole water sample after separation with an appropriate methodology. [Guidance CIRCA 19]

## WHOLE WATER MEASUREMENTS WHAT ARE WE TALKING ABOUT ?

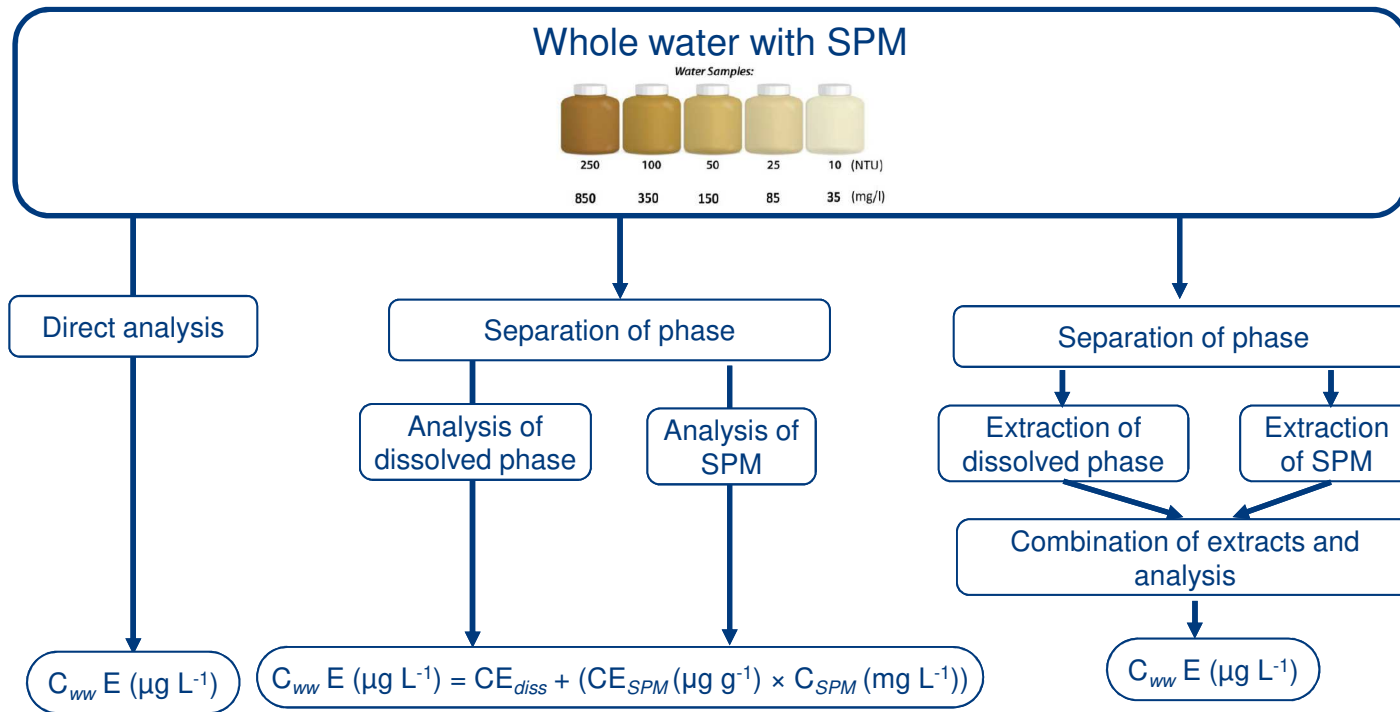
### Definitions

- **Total concentration of the analyte** shall mean the total concentration of the analyte in the whole water sample, reflecting both dissolved and particle bound concentrations of the analyte. [\[Guidance CIRCA 19\]](#)
- **Dissolved concentration of the analyte** shall mean the concentration of the analyte in the liquid (dissolved) fraction of a whole water sample. [\[Guidance CIRCA 19\]](#)
- **Particle bound concentration of the analyte** shall mean the concentration of the analyte bound to SPM. [\[Guidance CIRCA 19\]](#)



# WHOLE WATER MEASUREMENTS

## How to deal with Whole Water measurements ?



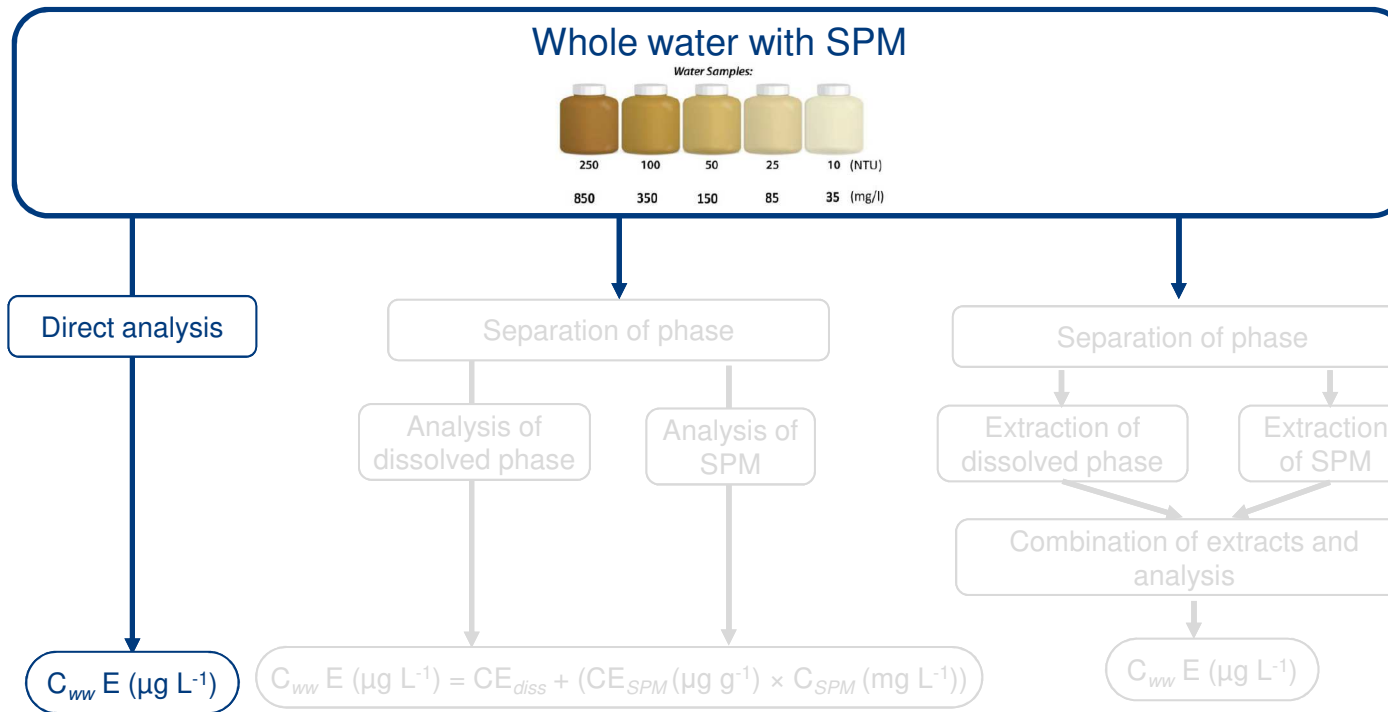
# WHOLE WATER MEASUREMENTS

## How to deal with Whole Water measurements

	Direct analysis	Separate analysis of SPM and Dissolved phase	Separate extraction of SPM and Dissolved phase and combined analysis
Advantages	<ul style="list-style-type: none"> <li>▪ Simple to implement</li> <li>▪ Limitations of intermediate steps in sample preparation to minimize risks of sorption and contamination</li> </ul>	<ul style="list-style-type: none"> <li>▪ Knowledge of the partitioning</li> <li>▪ Compatible with automated sample preparation approach</li> </ul>	<ul style="list-style-type: none"> <li>▪ Compatible with automated sample preparation approach</li> <li>▪ Overpass issue of calculation</li> </ul>
Limitations / Drawbacks	<ul style="list-style-type: none"> <li>▪ Absence of knowledge of the partitioning</li> <li>▪ Limitation on the choice of sample preparation approach</li> </ul>	<ul style="list-style-type: none"> <li>▪ Sufficient quantity of SPM, large volume sampling</li> <li>▪ Issue of calculation (impact of <math>C &lt; LOQ</math>)</li> <li>▪ Loss by sorption during the filtration</li> <li>▪ Risk of cross contamination during the filtration and multiple sample preparation approach</li> </ul>	<ul style="list-style-type: none"> <li>▪ Absence of knowledge of the partitioning</li> <li>▪ Loss by sorption during the filtration</li> <li>▪ Risk of cross contamination during the filtration and multiple sample preparation approach</li> <li>▪ Imply ad'hoc quantification approach</li> </ul>

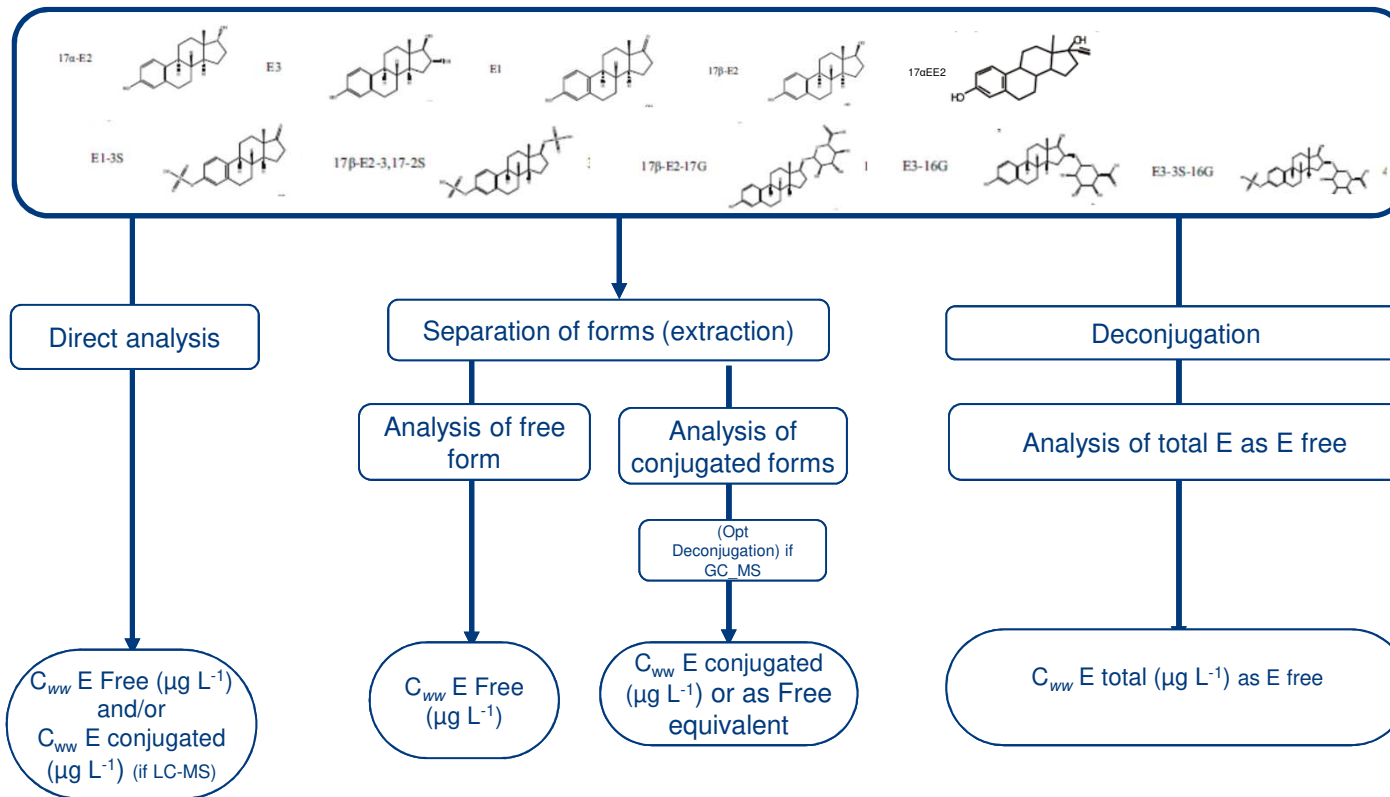
# WHOLE WATER MEASUREMENTS

## How to deal with Whole Water measurements



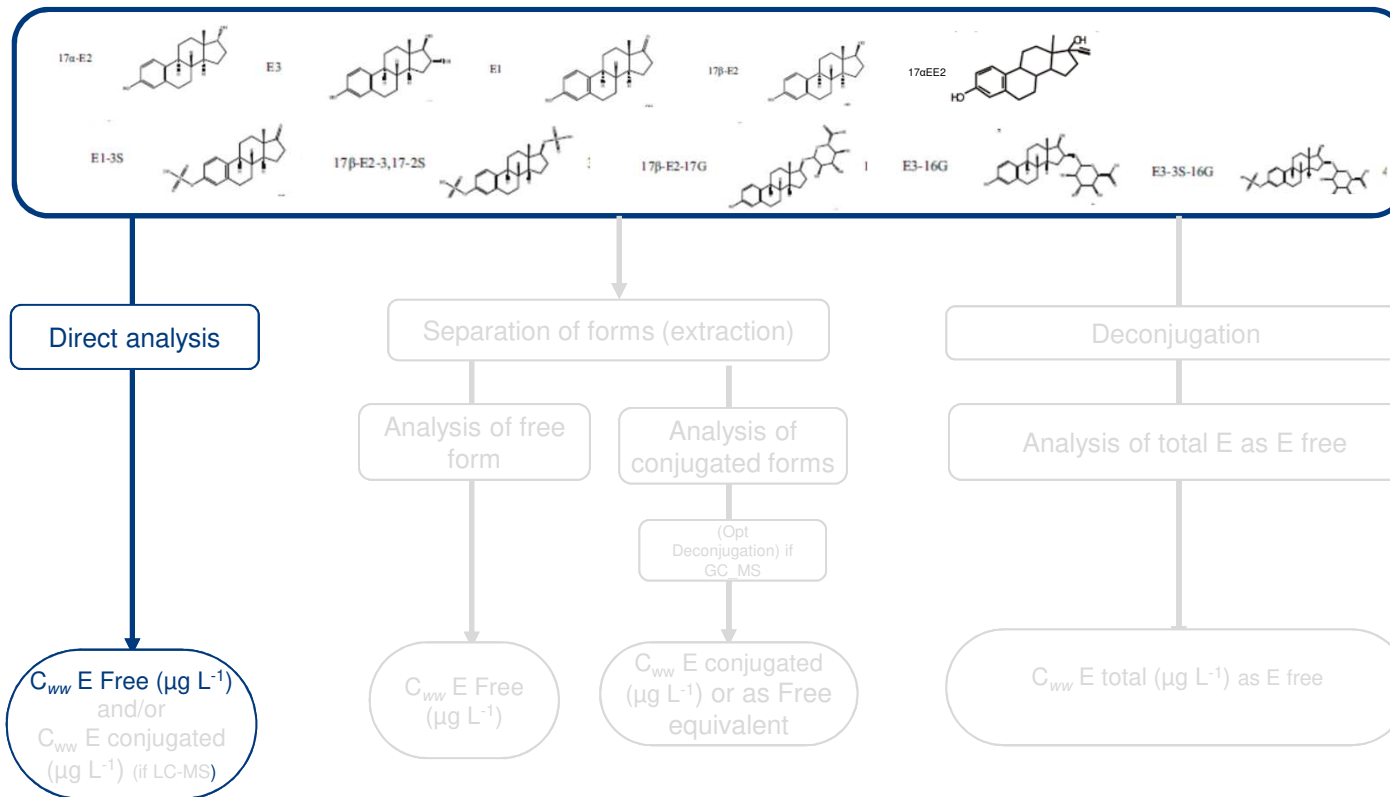
# WHOLE WATER ESTROGENS MEASUREMENTS

## How to deal with free/conjugates estrogens measurements



# WHOLE WATER ESTROGENS MEASUREMENTS

## How to deal with free/conjugates estrogens measurements



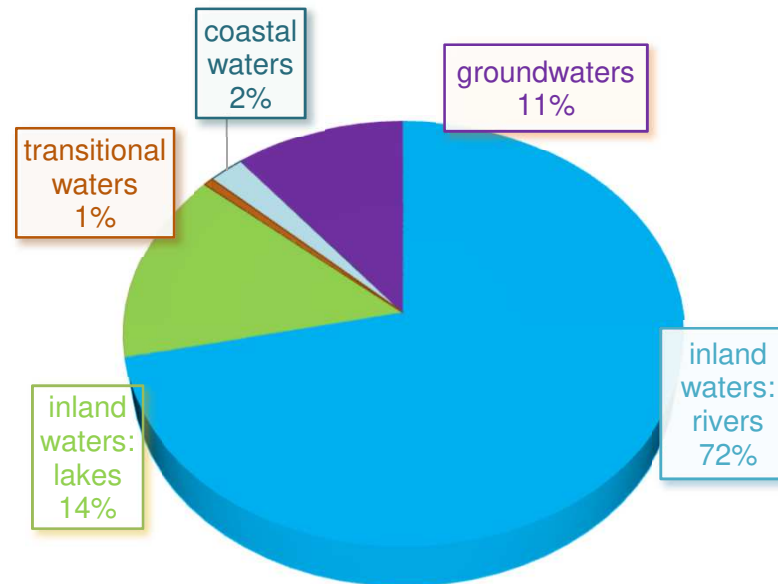
# EUROPEAN WATER BODIES

## Europe (25)

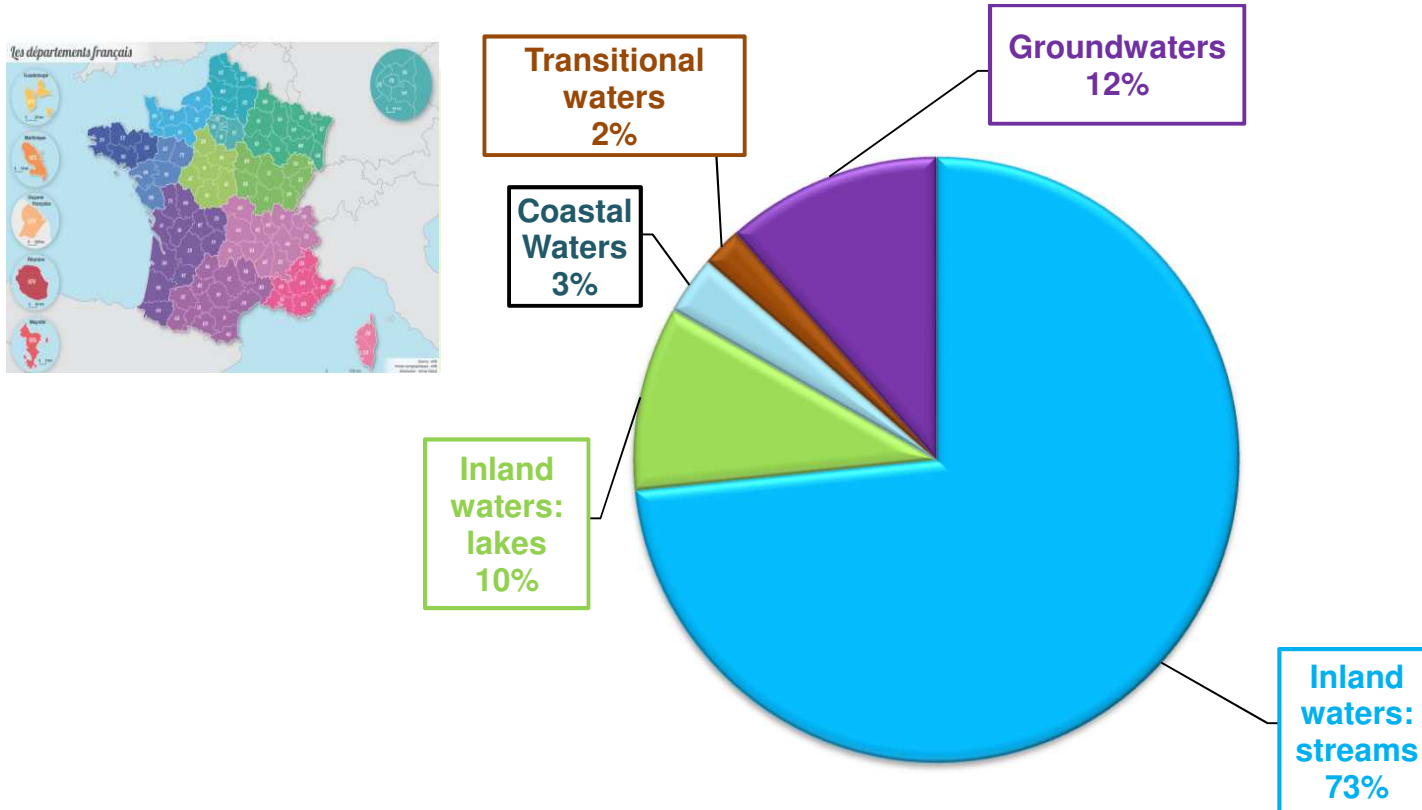


Source:  
<https://www.eea.europa.eu/themes/water/european-waters>

## TYPOLOGY OF WATER BODIES



# FRENCH WATER BODIES



Source FNE

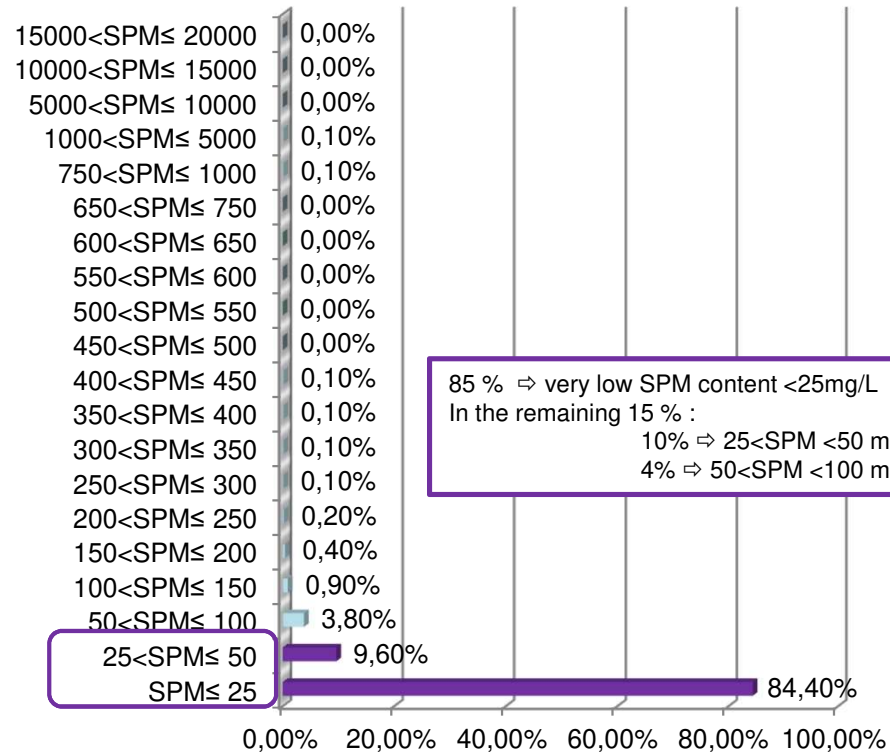
## Suspended Particulate Matter

### Inland surface waters

Min	0.06 mg/L
Max	16, 000 mg/L
Average	21.7 mg/L
Median	9 mg/L
Quartile 25	4.8 mg/L
Quartile 75	17 mg/L

Number of data = 188 283. according to EN 872 with few exceptions

Period : 03/01/2011 01/01/2018



Source :  
naiades.eaufrance.fr



➤ **Total Organic Carbon**

Min	0.09 mg C/L	Median	3.1 mg C/L
Max	190 mg C/L	Quartile 25	2.04 mg C/L
Average	3.80 mg C/L	Quartile 75	4.6 mg C/L

➤ **Conductivity**

Min	0.131 $\mu$ S/cm	Median	381 $\mu$ S/cm
Max	60,500 $\mu$ S/cm	Quartile 25	181 $\mu$ S/cm
Average	490 $\mu$ S/cm	Quartile 75	602 $\mu$ S/cm

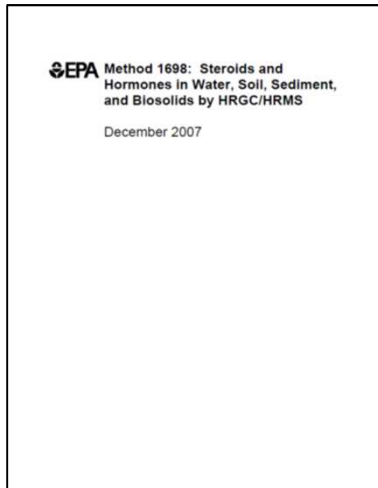
➤ **pH**

Min	5.3	Median	7.9
Max	12.96	Quartile 25	7.5
Average	7.88	Quartile 75	8.2

Source :  
naiades.eaufrance.fr

## AVAILABLE REFERENCES

### MS based methods



- Only one available CRM (17 $\beta$ E2) as primary calibrant to establish traceability to SI
- No matrix matched Certified Reference Material
- Some PT but material not representative (see eptis database)

## AVAILABLE REFERENCES

### EBM



#### ISO 19040-1

Water quality — Determination of the estrogenic potential of water and waste water — Part 1: Yeast estrogen screen (*Saccharomyces cerevisiae*)



#### ISO 19040-2

Water quality — Determination of the estrogenic potential of water and waste water — Part 2: Yeast estrogen screen (A-YES, *Arxula adenivorans*)



#### ISO 19040-3

Water quality — Determination of the estrogenic potential of water and waste water — Part 3: In vitro human cell-based reporter gene assay

- Only one available CRM ( $17\beta$  E2) as primary calibrant to establish traceability to SI
- No matrix matched Certified Reference Material
- Some PT (see eptis database) but design is not totally appropriate



EDC  WFD



**Training workshop: Solutions to tackle WFD requirements for estrogen determination in water**

7-9 september 2022

EDC  WFD

# SAMPLING

## SAMPLING

- ❑ Comply with general principles of ISO 5667 guidances
- ❑ Implement general principles for ultratraces analysis
- ❑ No specific risk of contamination on the field identified: environment, operator, ...

- ❑ Recommendations:



- use glass bottle + prevent from photodegradation (amber, green bottles, alumina foil, ...)
- avoid intermediate material to prevent from losses due to sorption and risks of cross contamination
- cleaning procedure should be implemented: eg calcination of glassware, rinsing, ...



Avoid plastics especially if EBM have to be implemented

# SAMPLING

## Illustration of sorption phenomena

Table 2. Relative mass lost over 24 h and first-order sorption rates for selected materials

Material	%E2†	%EE2	%E1
Type 304 stainless steel	24.9 ± 5.5	53.4 ± 2.0	52.7 ± 2.4
Type 316 stainless steel	30.5 ± 1.8	56.1 ± 3.7	54.3 ± 3.6
Glass (culture tubes)	1.0 ± 0.7	0.7 ± 0.6	0.6 ± 0.4
PolyCarbonate	8.7 ± 5.1	51.2 ± 8.3	44.1 ± 6.8
PVC‡	4.5 ± 2.5	5.0 ± 2.1	7.7 ± 4.5
Teflon	2.3 ± 1.3	4.2 ± 0.6	2.2 ± 0.8
Autoclaved			
Type 304 stainless steel	2.3 ± 7.0	33.7 ± 7.6	28.4 ± 6.1
Type 316 stainless steel	6.8 ± 17.7	33.2 ± 6.2	20.9 ± 5.8
Glass (culture tubes)	-4.9 ± 0.6	-0.2 ± 1.2	0.5 ± 0.8

- Minimum losses with glass container and materials

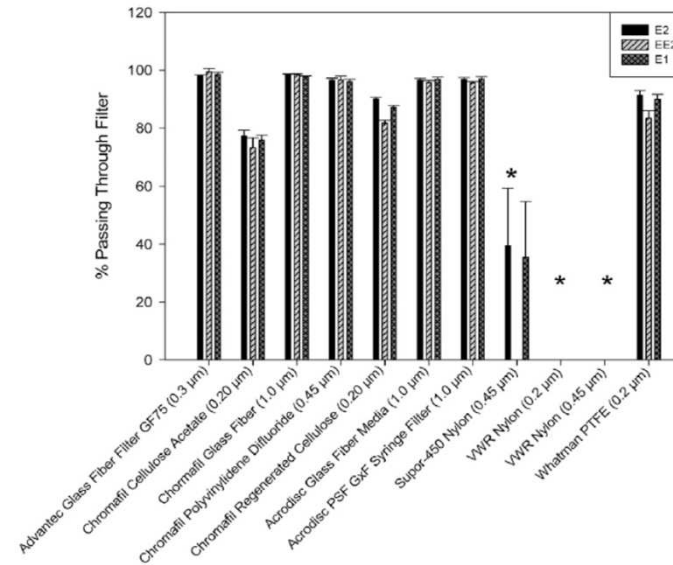


Fig. 3. Percent passing through different filter types. \*Significant differences ( $p < 0.05$ ) between the percent passing and the initial concentration. Error bars represent 1 SD of the mean for triplicate samples. E1, estrone; E2, 17 $\beta$ -estradiol; EE2, 17 $\alpha$ -ethynylestradiol.

Charles W. Walker\* and John E. Watson, 2010





# STABILITY/ INTEGRITY

# Stability

## Main drivers

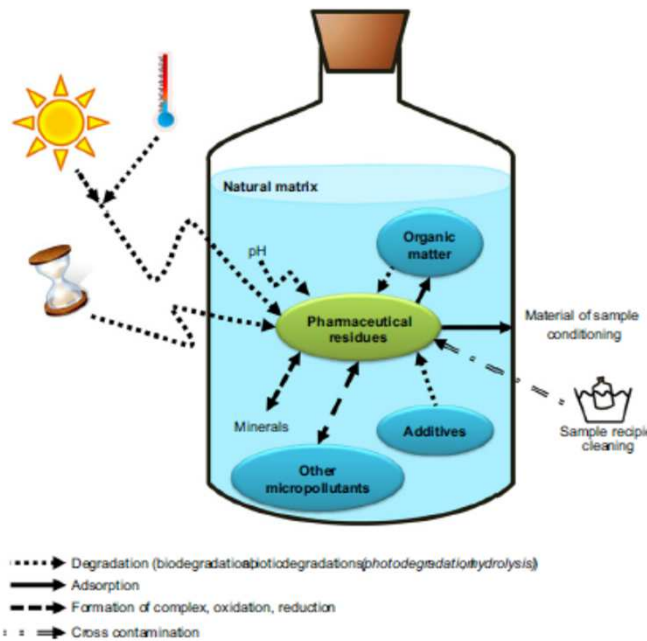


Fig. 1. Sources and processes (possibly) affecting the stability of PPs in samples before analysis.

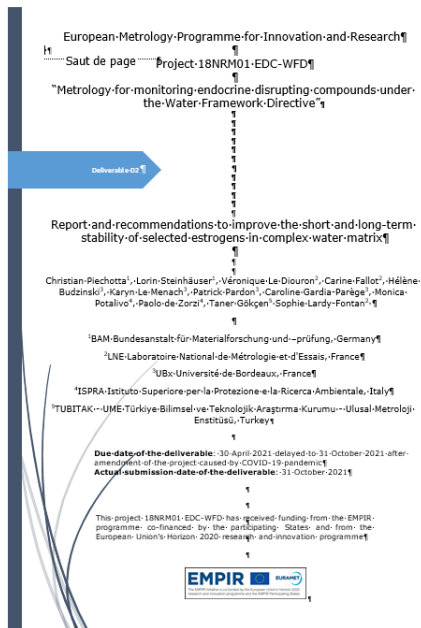
Numerous review and publications

⇒ Not all in agreement

- Methodological strategy
- Acceptance criteria
- Missing key information

Mompelat et al. (2013)

# Stability



## Following

- ISO GUIDE 35:2017 Reference materials — Guidance for characterization and assessment of homogeneity and stability  
⇒ develop candidate reference material(s)
- ISO TS 5667-25 Water quality — Sampling — Part 25: Guideline on the validation of the preservation time of water samples  
⇒ recommendations on maximum acceptable delay before analysis (MaxADs)

# Stability study design 1

## Stability water sample from sampling until analysis

- Five selected estrogens E1, 17 $\alpha$ E2, 17 $\beta$ E2, E3 and 17 $\alpha$ EE2 in water matrices
- Representative synthetic real water matrix: Evian water, DOC of 7 mg/L, pH 7.3 without SPM
- Concentration level of 10 ng/L for each species.
- Container: glassware+ protection from sunlight
- Sample preservation: : none, 0.1% methanol, 0.1% ascorbic acid
- Storage conditions: +4°C and room temperature
- Duration: 14 days
- Time laps : Day1, Day2, Day3, Day4, Day5, Day14
- Isochronous study : experimental study of “reference” material stability in which units exposed to different storage conditions and times are measured in a short period of time
  - Unstability at room temperature
  - Stability at 4°C up to 14 days
  - With Ascorbic acid ↗ SD%

## Stability study design 2

### Stability of a whole water sample from sampling until analysis

- five selected estrogens E1, 17 $\alpha$ -E2, 17 $\beta$ E2, E3 and 17 $\alpha$ -EE2
- Representative synthetic real water matrix: Evian water, DOC of 7 mg/L, pH 7.3 and SPM 50 mg/L
- Concentration level of 10 ng/L for each species.
- sample preservation: none, 0,1% methanol, (0,1% ascorbic acid)
- Container: glassware+ protection from sunlight
- Storage conditions: +4°C Duration: 14 days
- Time laps : Day1, Day 5, Day 14
- Isochronous study
  - without stabilising reagent or with 1% methanol show no significant trend with regards to degradation or loss of the analytes during storage time
  - With Ascorbic acid ↗ SD%

## Stability study design 3

### Stability of a whole water sample from sampling until analysis

- five selected estrogens E1, 17 $\alpha$ -E2, 17 $\beta$ E2, E3 and 17 $\alpha$ -EE2 in water matrices
  - Representative synthetic real water matrix: Evian water, DOC of 7 mg/L, pH 7.3 and SPM 50 mg/L iron and manganese oxidizing bacteria (*Sphingomonas spec. and Sphaerotilus spec.*)
  - Concentration level of 10 ng/L for each species.
  - Container: glassware+ protection from sunlight
  - Sample preservation: none, 0,1% ascorbic acid
  - Storage conditions: +4°C
  - Duration: 14 days
  - Time laps : Day1, Day 5, Day 14
  - Isochronous study
- impact of microbial activity showed only a small influence on the decrease of the estrogen concentration within a period of 14 days

# Stability

## ❑ Recommendations:



- Material in contact with water samples should be cleaned cautiously
- Glassware (coloured) shall be preferred to any other type of material if LOQ  $\ll 0,1$  ng/L targeted to minimize the risks of sorption/contamination (interferences)
- Storage at 4°C is OK for 7 days if low complex matrix.
- For complex matrix storage time shall be shortened. Addition of 1% MeOH is recommended to avoid biodegradation
- Storage at -20°C is possible but could be at risks of clogging if SPE cartridge are implemented. Some impacts on ME% (↗) have also been observed



Plastic container shall be avoided if LOQ  $\ll 0,1$  ng/L targeted





EDC  WFD



**Quantification strategies  
for estrogens with regards  
to the EU-WFD**

7-9 September 2022

EDC  WFD

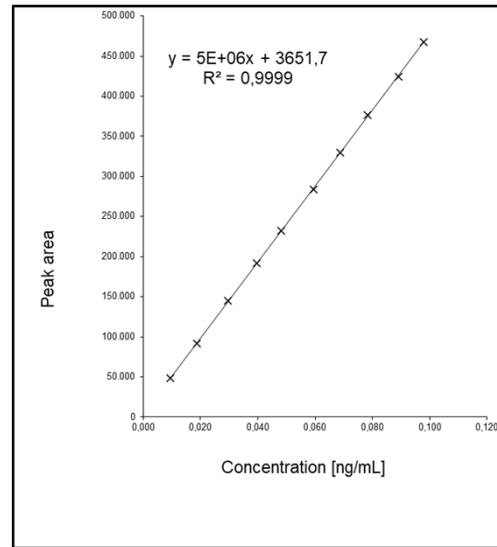
## Overview on the quantification of estrogens

### Selectivity – Identification of target estrogens in chromatograms with mass spec detection

- It is necessary to demonstrate that the identification of the target analytes is properly achieved and, moreover, that the target analytes signals are not influenced by the presence of chemically or physically interferences.
- Target analytes shall be identified in accordance with ISO 21253-1: Define target m/z ratios for each analyte (if possible a quantifier and at least two qualifier ions with distinct intensity ratios)
- Using tandem mass spec: implementation of transitions for the target molecules to specific fragments
- Using high res mass spec: define specific exact m/z ratios for the target molecules or of specific fragments

## Overview on the quantification of estrogens

- External calibration using unlabeled target estrogens:



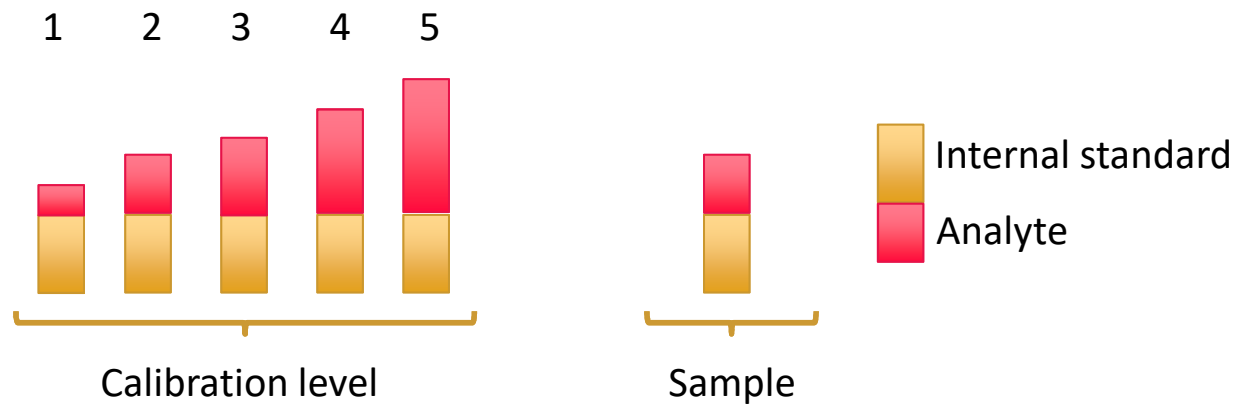
- After regression calculation of estrogen concentration by using the regression equation and the area of the sample extract obtained by LC- or GC-MS(/MS)

## Overview on the quantification of estrogens

- Loss of analyte during preconcentration and clean up steps is not considered
- Influence of matrix on the measurement can enhance or reduce the signal intensity compared to non-matrix containing calibration level
- Internal standard is needed to correct the loss of analyte during sample preparation or sample inlet to chromatography with mass spec detection
- Internal standard is a compound that is similar to chemical species of interest in the samples
- Physicochemical properties should be identically

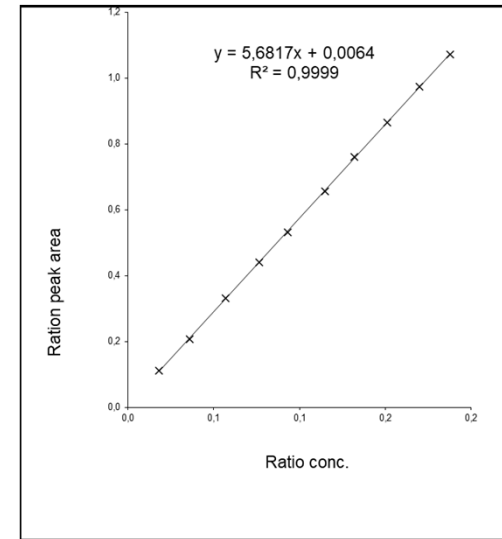
## Overview on the quantification of estrogens

- Internal standard of choice: stable isotopically labeled target analytes (e.g., deuterated,  $^{13}\text{C}$ - or  $^{18}\text{O}$ -labeled species)
- Identical boiling points, polarities, and fragmentation patterns in EI-MS
- Internal standard and analyte get the same impact by matrix, sample prep., purification and analysis



## Overview on the quantification of estrogens

	Concentration in ng L <sup>-1</sup>			peak area (counts)		Ratio peak area Analyt/INT
	Analyt	INT-STD	ratio conc. Analyt/INT	Analyt	INT	
Cal 1	0,098	0,521	0,19	467955	436874	1,07
Cal 2	0,089	0,524	0,17	424045	435361	0,97
Cal 3	0,078	0,517	0,15	376758	435804	0,86
Cal 4	0,069	0,522	0,13	330095	434732	0,76
Cal 5	0,060	0,519	0,11	283936	432834	0,66
Cal 6	0,048	0,516	0,09	231994	436468	0,53
Cal 7	0,040	0,521	0,08	191640	435065	0,44
Cal 8	0,029	0,520	0,06	145134	437598	0,33
Cal 9	0,019	0,524	0,04	91929	443525	0,21
Cal 10	0,010	0,524	0,02	48572	438664	0,11

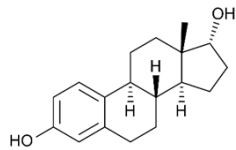


Solved equation from the linear regression:

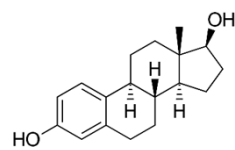
$$\beta_A = \frac{\left( \frac{y_A}{y_{IS}} - b \right)}{a} * \beta_{IS}$$

- Typical ten point calibration curve of estriol. Internal standard concentration at mid concentration level of the analyte
- Concentration ratio vs. area ratio obtained from the analysis

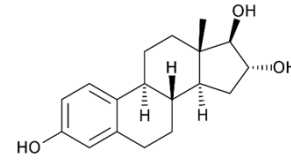
## Overview on the quantification of estrogens



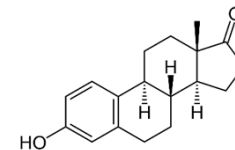
17-alpha-estradiol



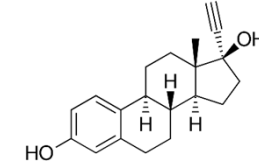
17-beta-estradiol



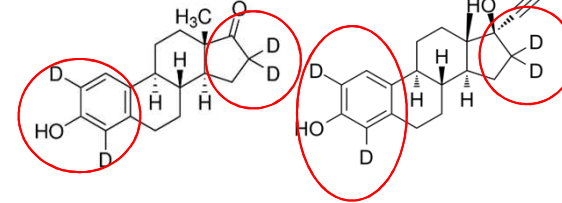
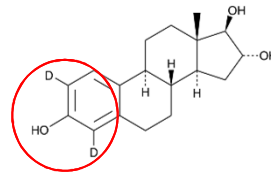
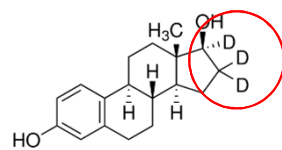
Estriol



Estrone



17-alpha-ethinylestradiol



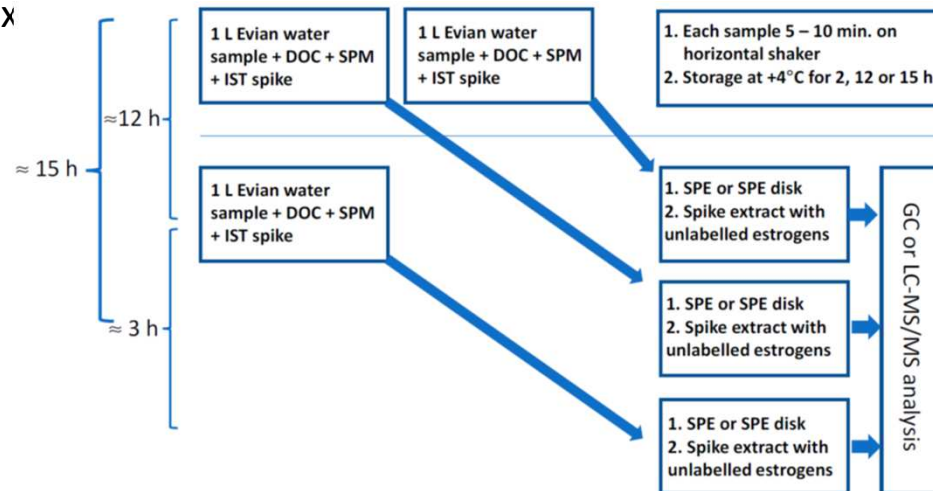
- In general  $^{13}\text{C}$ -labeled species are more stable than deuterated ones (deuterium-hydrogen exchange, depending on the position of the labeling)
- It could be demonstrated that deuterated estrogens are stable and can be used as  $^{13}\text{C}$ -labeled species (costs for routine labs)
- IDMS-technique – organic isotope dilution calibration



## Overview on the quantification of estrogens

### Equilibrium time for internal standards

- The presence of humic acid and SPM reduces the extraction efficiency down to 10 and 75% and the recovery of the internal standards depending on the equilibrium time. To obtain reliable estimates it is essential to allow a certain time for equilibration between the isotope spike and the aqueous matrix



## Overview on the quantification of estrogens

### Equilibrium time for internal standards

For the different labeled estrogens, an equilibrium time of at least 15 h is suggested. No difference is observed between the different types of internal standards (deuterated,  $^{13}\text{C}$ ). In consequence, the **samples** must be **spiked** with the **isotopically labeled standards** at the **end of a working day** to store them **overnight (15 h)** at **+4°C**.

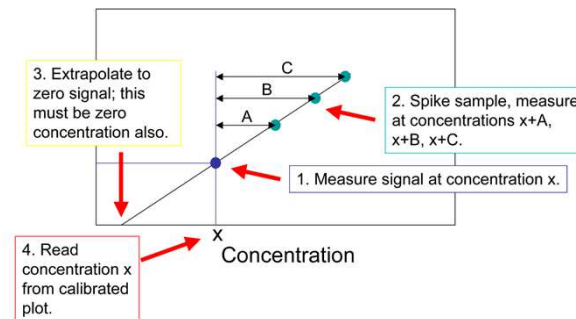
## Overview on the quantification of estrogens

- Within a typical method validation the method recovery (i.e. absolute and relative recovery) will be assessed so see the performance or capability of the analytical procedure.
- Absolute recovery will be determined in each matrix by comparing the peak areas of spiked samples prior and post extraction.
- Relative recoveries will be calculated in each matrix by comparing the ratios of the peak areas of the target analytes to the corresponding peak areas of the internal standards in spiked samples prior extraction and post extraction.

## Overview on the quantification of estrogens

### Alternative quantification strategies:

- Standard addition: a type of quantitative analysis approach whereby the standard is added directly to the analyzed sample. It is used where the sample matrix also contributes to the analytical signal, a situation known as the matrix effect, thus making it impossible to compare the analytical signal between sample and standard using the traditional calibration curve approach



- Use of non-isotopically labeled internal standard: Here, structural analogues were used e.g. 3-O-methyl estrone

## Overview on the quantification of estrogens

### Conclusion:

- Analyzing estrogens in complex water matrix suitable internal standards are needed
- To respect matrix effects and the loss of analytes during sample prep. and purification stable isotopically labeled species of the target analytes should be used
- Organic isotope dilution calibration with mass spectrometric detection is a powerful tool to increase the relative recovery of the target analytes within the analytical procedure



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7-9 September 2022

EDC  WFD



**Sample preparation  
strategies for estrogens  
with regards to the EU-  
WFD**

7-9 September 2022

EDC  WFD

## Overview on preconcentration procedures

### Needs for sample preparation

In general a compromise is necessary which addresses the whole water sample, a reasonable preconcentration factor, and the robustness of the method:

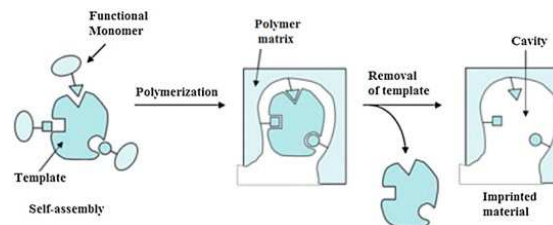
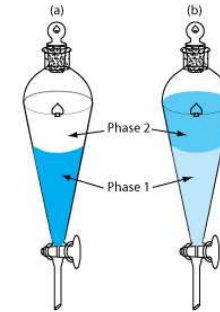
- Preconcentration of target analytes due to low EQS given by the EU-WFD
- Separation of matrix components from target analytes to prevent matrix effects on the chromatographic separation e.g., decreasing separation efficiency of the chromatographic column, enhancing or decreasing signal intensity (ion suppression in ESI source)
- Avoiding contamination of sample inlets (GC liner contamination by matrix components)

A preselection of feasible and applicable techniques was done by all project partner due to their practical knowledge



## Overview on preconcentration procedures

- Liquid-liquid extraction (LLE)
- Solid phase extraction (SPE) columns (off- and online)
- SPE disks
- Mi-SPE (molecular imprinted polymers)



## Overview on preconcentration procedures

### Liquid-liquid extraction (LLE)

- Use of high volumes of organic extraction solvents (commonly 30 – 50 mL; green chemistry)
- Only non-miscible organic solvents can be used for extraction (limited selectivity)
- Only “batch extraction”, must be repeated with portions of fresh extraction solvent
- Physical stressful for whom how do the extraction – it is not automatable – no extractions in parallel (not feasible for routine labs)
- Batch extracts must be combined and evaporated



**Not applicable in routine lab!**

## Overview on preconcentration procedures

### Liquid-liquid extraction (LLE)

Typical procedure:



500 mL water sample

- 3 x extraction with 30 mL DCM
- Extraction time 1 min



Combining organic layers

- Evaporation to dryness
- Reconstitution with 1 mL ACN



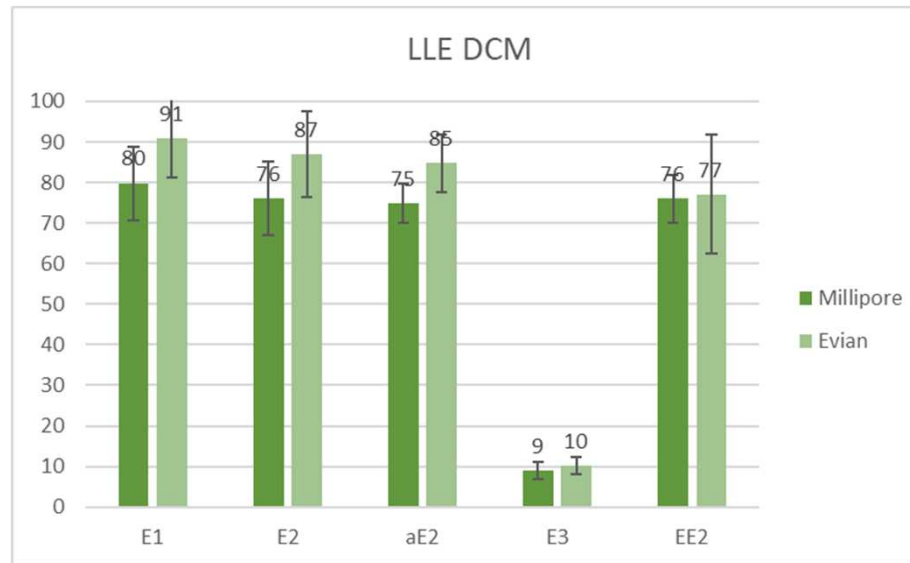
Analysis

- The ACN extract is ready for analysis



## Overview on preconcentration procedures

### Liquid-liquid extraction (LLE)



- Sufficient absolute recoveries for E1, E2, alphaE2 and EE2 but poor for E3 independently from the type of water matrix

## Overview on preconcentration procedures

### Solid phase extraction (SPE) off- or online

- E.g., HLB (hydrophilic-lipophilic balanced) 6 or 3 mL / filled with 100 or 60 mg sorbent material. Other materials are also feasible ( $C_{18}$ ,  $C_{18}$  eq, ...)
- With a content of less than  $50 \text{ mg L}^{-1}$  SPM no clogging can be observed (it strongly depends on the ratio of SPM to sample volume)
- Preconcentration not only of the target analytes but also enrichment of DOC content (was evaluated by measuring the DOC of the sample extract)
- Problems of ion suppression when doing LC-MS/MS (in most cases decrease of the signal intensity, known effect in the ESI source)



## Overview on preconcentration procedures

### A typical procedure



#### 1000 mL water sample

- Spiked with IS-mix at desired level
- 15 min on horizontal shaker
- Oasis HLB 3 mL/100mg



#### Conditioning / sample load

- Conditioning with 10 mL ACN
- Conditioning with 15 mL H<sub>2</sub>O
- Load 1000 mL water sample 20 mL/min



#### Washing step and elution

- Washing cartridge with 10 mL H<sub>2</sub>O
- Dry cartridge with N<sub>2</sub> for 1 min
- Collect 10 mL fraction into sample tube using ACN



#### Concentration and analysis

- Concentrate organic fraction to 1 mL (TurboVap)
- Extract is ready for analysis by LC-MS/MS



OASIS  
SAMPLE EXTRACTION PRODUCTS



## Overview on preconcentration procedures

### Solid phase extraction (SPE) off- or online

- Experiments were carried out by all project partners
- Different matrix compositions were used:
  - Low complex matrix (Evian water with low DOC and without SPM)
  - High complex matrix (Evian water with moderate DOC and with SPM)
- Relative and absolute recovery rates were determined to compare the different experiments with regards to their performance (applicability in routine analysis)

## Overview on preconcentration procedures

### Solid phase extraction (SPE) off- or online

Low complex matrix (Evian water with low DOC and without SPM)

Extraction method  Parameter	Waters Prime HLB size 3cc/60mg, Evian water (500ml) + DOC (7 mg/L) and estrogens at 5 ng/L level			
	Absolute recovery (%)	RSD (%)	Relative recovery (%)	RSD (%)
Estrone (E1)	111,8	5,0	101,3	0,6
17 $\alpha$ -Estradiol (aE2)	132,8	5,7	98,5	6,5
17 $\beta$ -Ethinylestradiol (bE2)	110,5	6,3	97,9	0,6
Estriol (E3)	118,4	6,9	101,0	5,9
17 $\alpha$ -Ethinylestradiol (EE2)	110,9	4,6	98,3	4,1

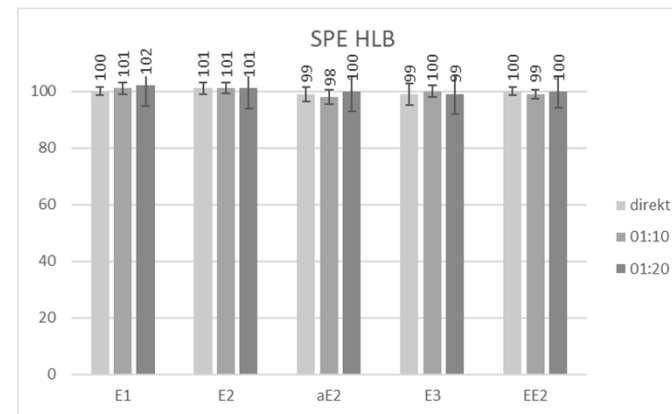
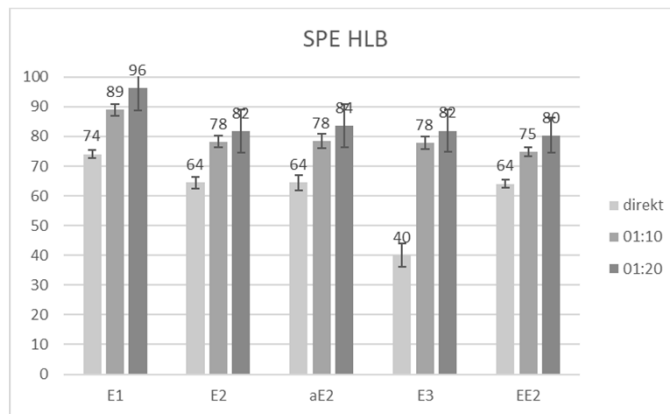
- Sufficient absolute recoveries for E1, E2, alphaE2, EE2 and E3
- Ideal relative recoveries by isotope dilution calibration
- But: no SPM in the water matrix which can cause clogging of the cartridge



## Overview on preconcentration procedures

### Solid phase extraction (SPE) off- or online

High complex matrix (Evian water with moderate DOC and with SPM)

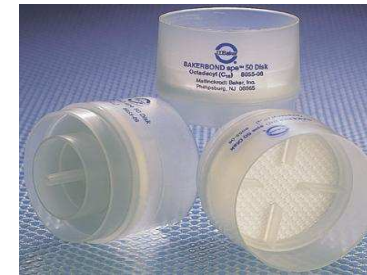


- Sufficient absolute recoveries for E1, E2, alphaE2, EE2 and E3
- Matrix will also be preconcentrated; increasing signals by diluting the sample extracts
- Ideal relative recoveries by isotope dilution calibration

## Overview on preconcentration procedures

### SPE disk (combination of filtration and SPE)

- Good compromise of filtration and enrichment
- E.g., HLB, DVB or C<sub>18</sub> disks with 47 mm diameter are feasible (capacity is only defined as L, H and M – not correlation to distinct amounts of SPE sorbents possible)
- Need of special manifold suitable for SPE disks
- Sufficient absolute recovery rates within 80 to 90% in complex water matrix (DOC, SPM and inorganic content)
- Applicable for SPM content higher than 50 mg L<sup>-1</sup> e.g., 500 mg L<sup>-1</sup>
- Preconcentration of DOC content comparable to common SPE



## Overview on preconcentration procedures

### A typical procedure



#### 1000 mL water sample

- Spiked with IS-mix at desired level
- 15 min on horizontal shaker
- SPE disk e.g., Atlantic HLB M/L or H



#### Conditioning / sample load

- Conditioning with 20 mL ACN
- Conditioning with 3 x 10 mL H<sub>2</sub>O
- Load 1000 mL water sample in approx. 30 min



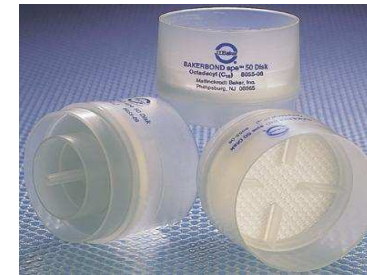
#### Washing step and elution

- Washing cartridge with 10 mL H<sub>2</sub>O
- Drying of SPE disk by applying vacuum for 5 min
- Collect 5 x 10 mL fractions using ACN



#### Concentration and analysis

- Concentrate organic fraction to 1 mL (TurboVap)
- Yellowish extract is ready for analysis by LC-MS/MS



## Overview on preconcentration procedures

### SPE disk (combination of filtration and SPE)

- Experiments were carried out by all project partners
- Different matrix compositions were used:
  - Low complex matrix (Evian water with low DOC and without SPM)
  - High complex matrix (Evian water with moderate DOC and with SPM)
- Relative and absolute recovery rates were determined to compare the different experiments with regards to their performance (applicability in routine analysis)

## Overview on preconcentration procedures

### SPE disk (combination of filtration and SPE)

Low complex matrix (Evian water with low DOC and without SPM)

Parameter	Extraction method			
	Absolute recovery (%)	RSD (%)	Relative recovery (%)	RSD (%)
Estrone (E1)	77,2	12,6	90,0	0,3
17 $\alpha$ -Estradiol (aE2)	NA	NA	86,8	0,4
17 $\beta$ -Ethinylestradiol (bE2)	88,1	16,3	84,5	10,0
Estriol (E3)	82,8	9,8	88,1	1,4
17 $\alpha$ -Ethinylestradiol (EE2)	72,6	10,2	93,8	5,6

- Sufficient absolute recoveries for E1, E2, alphaE2, EE2 and E3
- Ideal relative recoveries by isotope dilution calibration

## Overview on preconcentration procedures

### SPE disk (combination of filtration and SPE)

High complex matrix (Evian water with moderate DOC and with SPM)

Parameter	Extraction method			
	Absolute recovery (%)	RSD (%)	Relative recovery (%)	RSD (%)
Estrone (E1)	92,6	4,0	95,0	7,5
17 $\alpha$ -Estradiol (aE2)	92,4	8,8	88,1	6,5
17 $\beta$ -Ethinylestradiol (bE2)	90,8	9,7	88,1	1,4
Estriol (E3)	99,6	4,4	92,5	4,2
17 $\alpha$ -Ethinylestradiol (EE2)	97,1	10,7	99,0	2,6

- Sufficient absolute recoveries for E1, E2, alphaE2, EE2 and E3
- Ideal relative recoveries by isotope dilution calibration

## Overview on preconcentration procedures

### SPE disk (combination of filtration and SPE)

High complex matrix (Evian water with moderate DOC and with SPM)

Parameter	Extraction method			
	Absolute recovery (%)	RSD (%)	Relative recovery (%)	RSD (%)
Estrone (E1)	25,3	6,0	90,8	4,3
17 $\alpha$ -Estradiol (aE2)	NA	NA	78,4	9,7
17 $\beta$ -Ethinylestradiol (bE2)	29,4	4,8	90,5	5,0
Estriol (E3)	23,4	10,4	87,8	0,6
17 $\alpha$ -Ethinylestradiol (EE2)	20,7	4,4	104,5	2,6

- Poor absolute recoveries for E1, E2, alphaE2, EE2 and E3
- Ideal relative recoveries by isotope dilution calibration

# Overview on preconcentration procedures

AFFINIMIP® SPE

Estrogens

## Mi-SPE (molecular imprinted polymers)

### PROTOCOL OF PURIFICATION

#### Sample preparation

100mL of tap water spiked with 17β-E2-d<sub>3</sub> to a final concentration of 75ng/L was the loading solution.

Purification with a 3mL/100mg AFFINIMIP® SPE Estrogens cartridge

#### Equilibration

- 3mL Acetonitrile
- 3mL Water

#### Loading solution from sample preparation

#### Washing of interferents

- 3mL water
- 3mL Water/Acetonitrile (60/40)

#### Elution (E)

3mL Methanol

### Publications

Data extracted from **Determination of steroidal oestrogens in tap water samples using solid-phase extraction on a molecularly imprinted polymer sorbent and quantification with gas chromatography-mass spectrometry (GC-MS)**, D. Zacs, I. Perkons, V. Bartkevics, *Environ Monit Assess* 188, 433, 2016.

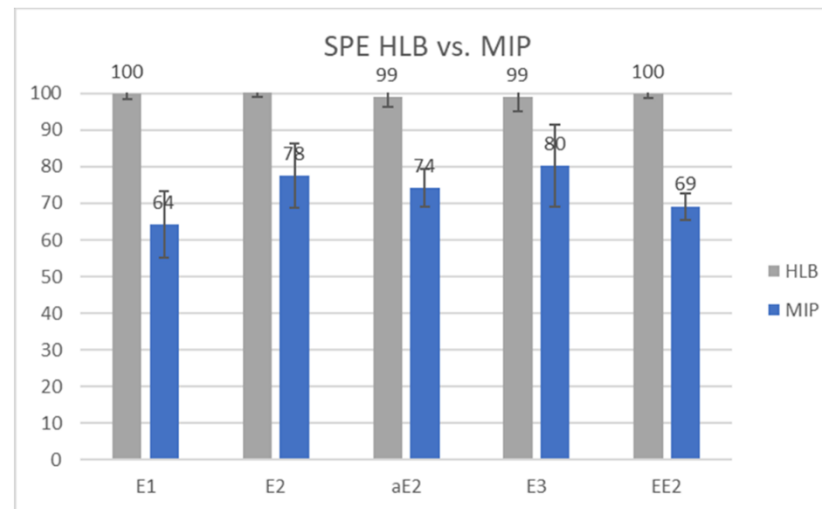
- Given protocol by the cartridge provider – limited sample volume of 100 mL – not suitable for estrogens with regards to EU-WFD
- Advantage: specific preconcentration of estrogens – key lock mode of action
- Ideal procedure to separate matrix (e.g., DOC, inorganics, ....) from estrogens
- Possible implementation as purification procedure for extracts obtained from preconcentration steps



## Overview on preconcentration procedures

### Mi-SPE (molecular imprinted polymers)

Low complex matrix (Evian water with low DOC and without SPM)



- Moderate relative recoveries for E1, E2, alphaE2, EE2 and E3 in comparison to common SPE; no high complex matrix experiments possible due to clogging of the cartridge

## Conclusions on preconcentration procedures

Overview of preconcentration methods including limitations and restrictions  
(O: fulfilling , +: good, ++: excellent, -: poor, --: unsatisfactory)

Extraction method Parameter	DI	SPE		SPE disk	LLE	Mi-SPE
		off-	online			
Preconcentration	--	++	++	++	O	--
Clean-up	--	+	+	+	O	++
Compatible to common solvents	--	+	+	+	-	O
Selectivity	--	++	++	++	O	++
Time / efficiency	++	+	++	++	+	O

## Conclusions on preconcentration procedures

Parameter \ Extraction method (Typical preconcentration factor)	SPE (1:1000)		SPE disk (1:1000)		MiSPE (1:100)	
	1	2	1	2	1	2
	Estrone (E1)	Green	Green	Green	Green	Yellow
17 $\alpha$ -Estradiol (aE2)	Green	Green	Green	Green	Yellow	Red
17 $\beta$ -Ethinylestradiol (bE2)	Green	Green	Green	Green	Yellow	Red
Estriol (E3)	Green	Green	Green	Green	Yellow	Red
17 $\alpha$ -Ethinylestradiol (EE2)	Green	Green	Green	Green	Yellow	Red

- Focusing only on the solid phase extraction techniques SPE is applicable for whole water samples with a SPM load less than 50 mg L<sup>-1</sup> to avoid clogging of the cartridge
- For higher SPM load the SPE disk is favored from all project partners as the method of choice

## Overview on purification/ clean up procedures

### Mi-SPE (molecular imprinted polymers)

#### PROTOCOL OF PURIFICATION

##### Sample preparation

100mL of tap water spiked with 17 $\beta$ -E2-d<sub>3</sub> to a final concentration of 75ng/L was the loading solution.

Purification with a 3mL/100mg AFFINIMIP<sup>®</sup> SPE Estrogens cartridge

##### Equilibration

- 3mL Acetonitrile
- 3mL Water

##### Loading solution from sample preparation

##### Washing of interferences

- 3mL water
- 3mL Water/Acetonitrile (60/40)

##### Elution (E)

3mL Methanol

- Extract from preconcentration step can be evaporated to at least 1 mL and can be diluted using MilliQ water up to 100 mL
- The volume is compatible to the max volume given by the provider SOP
- Advantage: specific preconcentration of estrogens – key lock mode of action
- Ideal procedure to separate matrix (e.g., DOC, inorganics, ....) from estrogens
- DOC in the Mi-SPE is decreased to a minimum (color changed from yellowish to a complete clear and colorless extract (evaluated by complementary DOC measurements) with sufficient recoveries

## Overview on purification/ clean up procedures

### Other SPE cartridges for purification of sample extracts

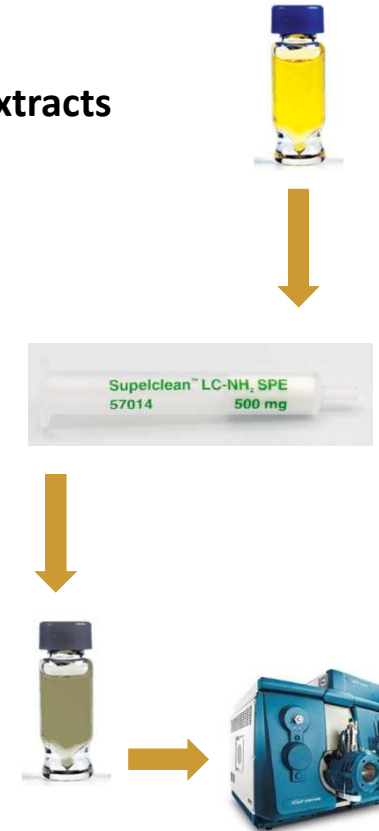
- For an alternative clean up of the sample extracts a Supelclean™ LC-NH<sub>2</sub> SPE (500 mg, 6 mL) cartridge (Merck, Darmstadt, Germany) can be used
- Silica gel based material with amino bounding functional groups
- Increase of matrix components but not specific as MiSPE



- Also mixed phase cartridges containing e.g., C<sub>18</sub> and aminopropyl functionality are available (e.g., Chromabond NH<sub>2</sub>/C<sub>18</sub>). Due to limited capacity not sufficient as a two step procedure

## Overview on purification/ clean up procedures

### Other SPE cartridges for purification of sample extracts



## Conclusions

- Appropriate techniques either for preconcentration or for purification of the extracts could be evaluated and established within the project by the partners
- Using e.g., HLB sorbent materials (SPE or SPE disk) the DOC will be co-preconcentrated
- Only 50 to 70% of the initial DOC can be removed by the preconcentration
- This can interfere the analytical method e.g., in case of LC-ESI-MS ion suppression can occur
- The result: loss of sensitivity
- A further purification of the extract from the preconcentration procedure is recommended by the project consortium

## Two-step procedure preconcentration/purification

- Experiments were carried out by all project partners
- High complex matrix (Evian water with moderate DOC and with SPM)
- Relative and absolute recovery rates were determined to compare the different experiments with regards to their performance (applicability in routine analysis)



## Two-step procedure preconcentration/purification

### SPE preconcentration + SPE purification

High complex matrix (Evian water with moderate DOC and with SPM)

#### SPE – LC-NH<sub>2</sub> SPE purification

Extraction method  Parameter	SPE extraction OASIS <b>HLB 500mg/6cc</b> + SPE purification <b>LC-NH2 500mg/3cc, 200 mL</b> Evian water + SPM (50mg/L) + DOC (7 mg/L) and estrogens at <b>0,4 ng/L level excepted 17<math>\alpha</math>-ethynylestradiol at 0,035ng/L</b> . Sample acidification at pH = 5, 1% MeOH added.			
	absolute recovery in %	RSD in %	Isotope dilution recovery in %	RSD in %
Estrone (E1)			98	2
17 $\alpha$ -Estradiol (aE2)			98	15
17 $\beta$ -Estradiol (bE2)			108	11
Estriol (E3)			116	7
17 $\alpha$ -Ethinylestradiol (EE2)			109	10
Estrone-13C3 (E1-13C3)	83	7		
17 $\alpha$ -Estradiol-d2 (aE2-d2)	74	4		
17 $\beta$ -Estradiol-13C3 (bE2-13C3)	81	6		
Estriol-d2 (E3-d2)	80	8		
17 $\alpha$ -Ethinylestradiol-d4 (EE2-d4)	79	5		

## Two-step procedure preconcentration/purification

### SPE disk preconcentration + SPE purification

High complex matrix (Evian water with moderate DOC and with SPM)

#### SPE disk – LC-NH<sub>2</sub> SPE purification

Estrogen	Evian® + SPM (50 mg L <sup>-1</sup> )	Evian® + DOC (5 mg L <sup>-1</sup> )	Evian® + DOC (5 mg L <sup>-1</sup> ) + SPM (150 mg L <sup>-1</sup> )
EE2	68 ± 22%	67 ± 3%	88 ± 1%
E3	64 ± 14%	62 ± 6%	76 ± 4%
aE2	68 ± 20%	70 ± 2%	85 ± 2%
bE2	67 ± 19%	69 ± 9%	90 ± 2%
E1	68 ± 22%	66 ± 1%	91 ± 4%

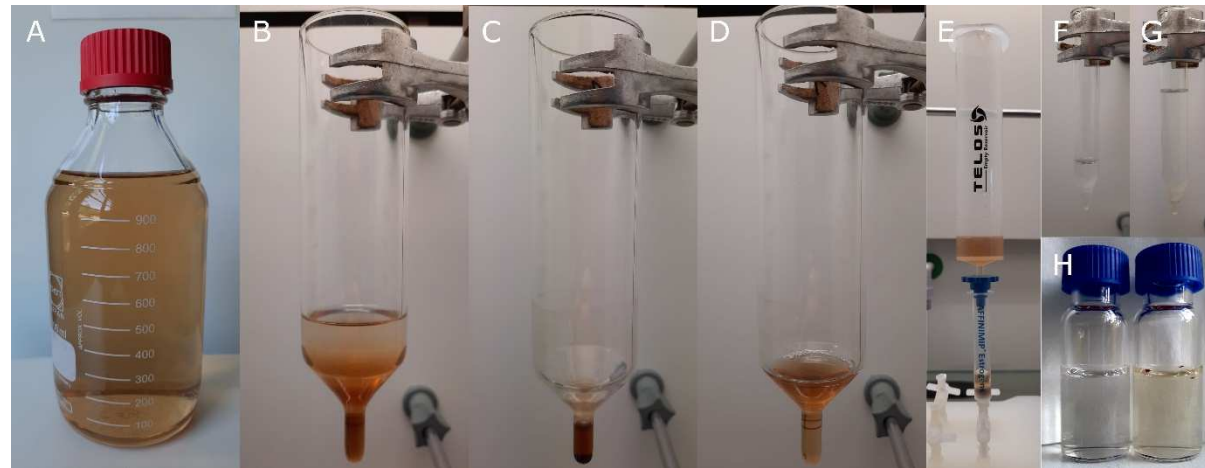
Estrogen	Evian® + SPM (50 mg L <sup>-1</sup> ) (%)		Evian® + DOC (5 mg L <sup>-1</sup> ) (%)		Evian® +DOC (5 mg L <sup>-1</sup> ) SPM (150 mg L <sup>-1</sup> ) (%)	
EE2	109	5	93	5	84	1
E3	97	8	90	9	86	6
aE2	98	1	99	4	98	0,40
bE2	96	2	93	4	92	4
E1	105	1	98	2	96	0,50

## Two-step procedure preconcentration/purification

### SPE disk preconcentration + SPE purification

High complex matrix (Evian water with moderate DOC and with SPM)

#### SPE disk – MiSPE purification



- DOC content of the preconcentrated and purified sample is less than 1% than the initial one (picture H left)

## Two-step procedure preconcentration/purification

### SPE disk preconcentration + SPE purification

High complex matrix (Evian water with moderate DOC and with SPM)

#### SPE disk – MiSPE purification

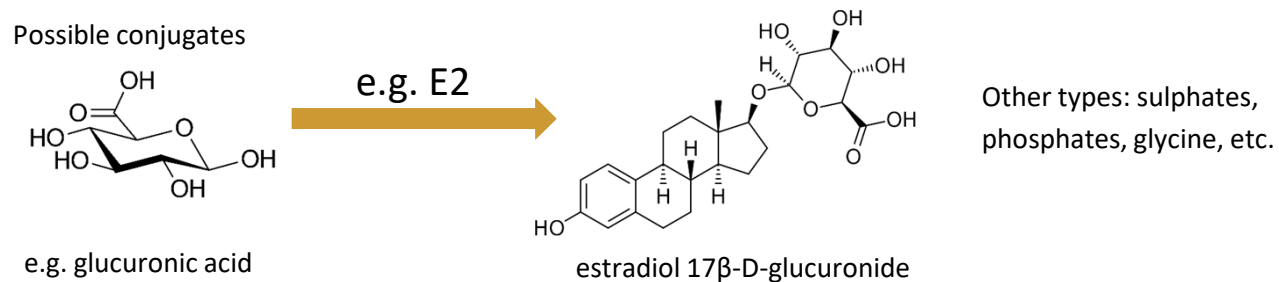
1 L Evian + 7 mg L<sup>-1</sup> DOC and 50 mg SPM spiked with estrogens at 0.1 ng L<sup>-1</sup>

	HLB disk		MiSPE		HLB disk + MiSPE	
	Relative recovery [%]	Standard deviation [%]	Relative recovery [%]	Standard deviation [%]	Relative recovery [%]	Standard deviation [%]
Estrone	97.00	0.25	92.54	0.105	96.20	0.30
Estradiol	100.02	0.01	100.02	0.01	100.01	0.01
17 $\alpha$ -Estradiol	100.09	0.02	100.07	0.04	100.14	0.06
Estriol	99.65	0.09	99.16	0.21	99.60	0.07
Ethinylestradiol	100.17	0.05	100.21	0.06	100.12	0.04

## Conjugates in whole water samples

### Conjugated target estrogens (optional)

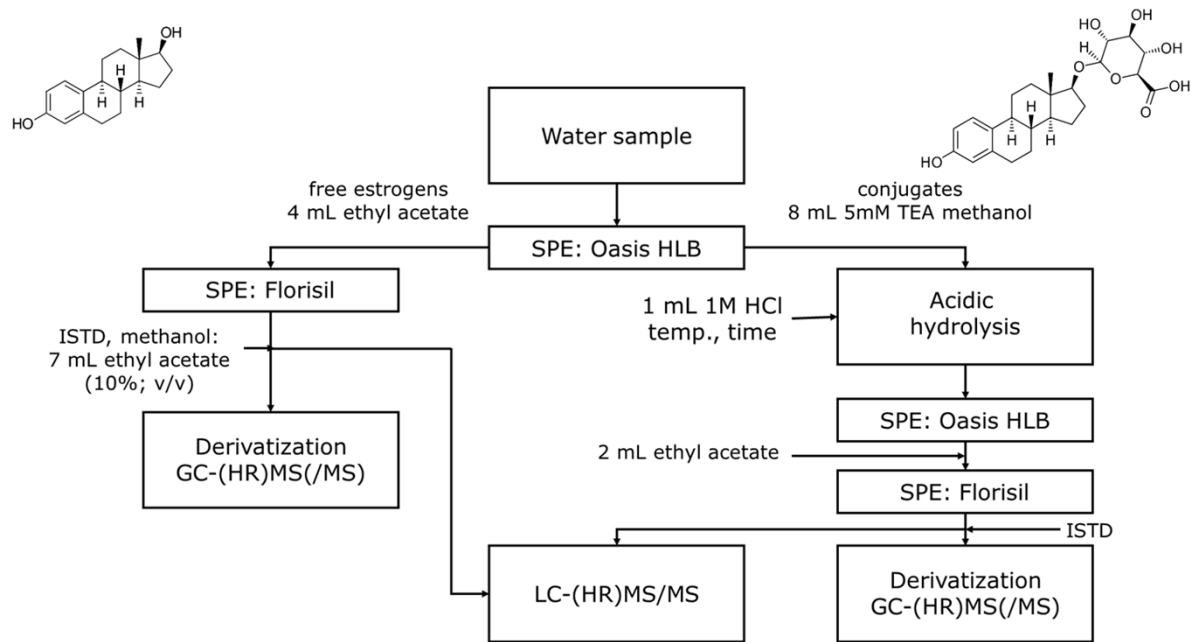
- Further phase-II-metabolism of E1, E2 and E3 leading to conjugated species
- Conjugates can be reconverted back into estrogens as a kind of a circulating reservoir for estrogens



- To be known: Influence of the sampling, sample storage, and sample preparation on the conjugates stability

# Conjugates in whole water samples

## Conjugated target estrogens (optional)



According to Liu, Z.-h., et al. (2010). "Deconjugation characteristics of natural estrogen conjugates by acid-catalyzed solvolysis and its application for wastewater samples." *Journal of Environmental Monitoring* **12(8): 1594-1600.**



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7-9 September 2022