

INTERNATIONAL WORKSHOP

Metrology for sustainable hydrogen energy applications

Impurities found in real hydrogen production samples

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Probability of impurity presence from expert industrial knowledge

Rationale of impurities presence based on production process and following an approach in ISO 19880.

Probability of impurity presence	Steam methane reforming with PSA	PEM water electrolysis process with TSA	Chlor-alkali process (membrane cell process)
Frequent	CO	None identified	O ₂
Possible	N ₂	None identified	None identified
Rare	Ar, CH ₄	N ₂ , O ₂ , H ₂ O	H ₂ O, N ₂
Very Rare	HCHO	CO ₂	CO ₂
Unlikely	He, O ₂ , CO ₂ , HCOOH, NH ₃ , sulfur compounds, hydrocarbons (except CH ₄), halogenated compounds	He, Ar, CO, CH ₄ , HCHO, HCOOH, NH ₃ , sulfur compounds, hydrocarbons (except CH ₄), halogenated compounds	He, Ar, CO, CH ₄ , HCHO, HCOOH, NH ₃ , sulfur compounds, hydrocarbons (except CH ₄), halogenated compounds

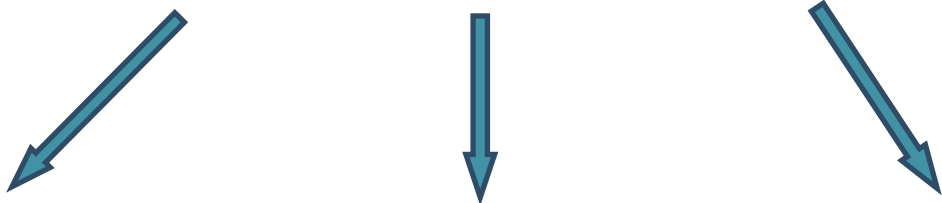
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Probability of occurrence of ISO 14687-2 contaminants in hydrogen: Principles and examples from steam methane reforming and electrolysis (water and chlor-alkali) production processes model

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Hydrogen purity measurements according to ISO 14687-2 and risk assessment for fuel cells



Probability of impurity presence from expert industrial knowledge



Sampling hydrogen on sites and analysis campaign at NMIs



Impact of trace concentration impurities on PEMFC performance



For hydrogen fuel quality and fuel quality control compliance
ISO 14687-2 and ISO 19880-8

Analytical campaign for hydrogen production methods

Analytical method used by National Metrology Institutes

Compounds	ISO 14687-2 threshold [μmol/mol]	NPL National Physical Laboratory	VSL Dutch Metrology Institute	RISE Research Institutes of Sweden	CEM CENTRO ESPAÑOL DE METROLOGÍA
Water H ₂ O	5	Quartz crystal microbalance	-	-	-
Methane CH ₄	2	GC-methaniser-FID	-	GC-FID	-
Non methane hydrocarbons	2	GC-methaniser-FID	-	GC-FID	-
Oxygen O ₂	5	GC-PDHID	-	GC-TCD	GC-TCD
Helium He	300	-	-	-	GC-TCD
Nitrogen N ₂	100	GC-PDHID	-	GC-TCD	GC-TCD
Argon Ar	100	GC-PDHID	-	GC-TCD	GC-TCD
Carbon dioxide CO ₂	2	GC-methaniser-FID	-	OFCEAS	-
Carbon monoxide CO	0.2	GC-methaniser-FID	-	OFCEAS	-
Total sulphur compounds	0.004	GC-SCD	-	OFCEAS	-
Formaldehyde HCHO	0.01	-	CRDS	-	-
Formic acid HCOOH	0.2	-	CRDS	-	-
Ammonia NH ₃	0.1	-	CRDS	-	-
Total halogenated (HCl)	0.05	-	CRDS	-	-
C2 hydrocarbons	2	-	-	TD-GC-FID/MS	-
C3 hydrocarbons	2	-	-	TD-GC-FID/MS	-
C4 hydrocarbons	2	-	-	TD-GC-FID/MS	-
C5 hydrocarbons	2	-	-	TD-GC-FID/MS	-
C6 - C18 hydrocarbons	2	-	-	TD-GC-FID/MS	-

GC: gas chromatography

PDHID: Pulse discharge helium ionisation detector

FID: Flame ionisation detector

CRDS: cavity ring down spectroscopy

TCD: Thermal Conductivity Detector

OFCEAS: Optical Feedback cavity enhanced adsorption spectroscopy

TD: Thermo-Desorption

MS: Mass spectrometry

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Hydrogen fuel quality from production methods (1)

Analytical campaign – Summary – Hydrogen production methods

Compounds	ISO 14687-2 threshold [μmol/mol]	SMR with PSA	PEM water electrolysis with TSA
		(5 samples) Results [μmol/mol]	(7 samples) Results [μmol/mol]
Water H ₂ O	5	< 0.5 – 2.5	< 3
Methane CH ₄	2	< 0.02 – 0.05	< 0.02
Non methane hydrocarbons	2	< 0.05	0.08 – 0.2
Oxygen O ₂	5	< 0.5	< 0.5 - 2
Helium He	300	< 50	< 9 - 45
Nitrogen N ₂	100	< 1.2 - 2	< 1.0 – 4.6
Argon Ar	100	< 0.5	< 0.5
Carbon dioxide CO ₂	2	< 0.02 – 0.45	< 0.02 – 0.25
Carbon monoxide CO	0.2	< 0.02	< 0.02
Total sulphur compounds	0.004	< 0.0036	< 0.0036
Formaldehyde HCHO	0.01	< 0.005	< 0.005
Formic acid HCOOH	0.2	< 0.1	< 0.1
Ammonia NH ₃	0.1	< 0.1	< 0.1
Total halogenated	0.05	< 0.005	< 0.005
C2 hydrocarbons	2	< 0.5	< 0.5
C3 hydrocarbons	2	<1	<1
C4 hydrocarbons	2	<1	<1
C5 hydrocarbons	2	<1	<1
C6 - C18 hydrocarbons	2	<0.05	<0.05

Hydrogen fuel quality from production methods (2)

Analytical campaign – Summary – PEM water electrolysis

Compounds	ISO 14687-2 threshold [μmol/mol]	PEM water electrolysis with TSA	PEM water electrolysis
		(7 samples) Results [μmol/mol]	(5 samples) Results [μmol/mol]
Water H ₂ O	5	< 3	> 100
Methane CH ₄	2	< 0.02	< 0.02 – 0.1
Non CH ₄ hydrocarbons	2	0.08 – 0.2	< 0.02 – 0.09
Oxygen O ₂	5	< 0.5 - 2	18- > 500
Helium He	300	< 9 - 45	< 9
Nitrogen N ₂	100	< 1.0 – 4.6	1.2 – 4.5
Argon Ar	100	< 0.5	< 0.5
Carbon dioxide CO ₂	2	< 0.02 – 0.25	0.2 – 5.4
Carbon monoxide CO	0.2	< 0.02	< 0.02
Total sulphur compounds	0.004	< 0.0036	< 0.0036
Formaldehyde HCHO	0.01	< 0.005	< 0.005
Formic acid HCOOH	0.2	< 0.1	< 0.1
Ammonia NH ₃	0.1	< 0.1	< 0.1
Total halogenated	0.05	< 0.005	< 0.005
C2 hydrocarbons	2	< 0.5	< 0.5
C3 hydrocarbons	2	<1	<1
C4 hydrocarbons	2	<1	<1
C5 hydrocarbons	2	<1	<1
C6 - C18 hydrocarbons	2	<0.05	<0.05

Steam methane reforming with pressure swing adsorption - Analysis results

		ISO 14687-2 threshold	Sample 1	Sample 2	Sample 3	Sample 4	Sample 5
Contaminants	Unit		Results with expanded uncertainty (k=2)				
CO	µmol/mol	0.2	< 0.01	< 0.01	< 0.01	< 0.053	< 0.053
CO ₂	µmol/mol	2	< 0.01	< 0.01	< 0.01	0.042 ± 0.016	< 0.02
CH ₄	µmol/mol	2	< 0.01	< 0.01	< 0.01	0.044 ± 0.007	< 0.02
Non methane hydrocarbons	µmol/mol	2	< 0.01	< 0.01	< 0.01	< 0.05	< 0.05
H ₂ O	µmol/mol	5	< 0.5	< 0.5	< 0.5	< 0.6	< 0.6
Total sulphur compounds	µmol/mol	0.004	< 0.0036	< 0.0036	< 0.0036	< 0.002	< 0.002
O ₂	µmol/mol	5	< 0.5	< 0.5	< 0.5	0.39 ± 0.13	0.39 ± 0.13
N ₂	µmol/mol	100	< 1.2	< 1.2	< 1.2	1.5 ± 0.6	< 1.0
Ar	µmol/mol	100	< 0.5	< 0.5	< 0.5	2.8 ± 0.1	< 0.5
Total halogenated (HCl)	µmol/mol	0.05	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O	µmol/mol	0.01	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O ₂	µmol/mol	0.2	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1
NH ₃	µmol/mol	0.1	n.m.	< 0.1	< 0.1	< 0.1	< 0.1
He	µmol/mol	300	n.m.	n.m.	< 50	20 ± 4	12 ± 5
C2 hydrocarbons	µmol/mol	2	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5
C3-hydrocarbons	µmol/mol	2	<1	<1	<1	<1	<1
C4-hydrocarbons	µmol/mol	2	<1	<1	<1	<1	<1
C5-hydrocarbons	µmol/mol	2	<1	<1	<1	<1	<1
C6 – C18 hydrocarbons	µmol/mol	2	<0.050	<0.050	<0.050	<0.050	<0.050

PEM water electrolyser with temperature swing adsorption - Analysis results

		ISO 14687-2 threshold	Sample 1	Sample 2	Sample 3	Sample 4	Sample 5	Sample 6	Sample 7	Sample 8
	Unit		Results with expanded uncertainty (k=2)							
CO	µmol/mol	0.2	< 0.053	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02	< 0.02
CO ₂	µmol/mol	2	0.443 ± 0.010	0.245 ± 0.010	0.229 ±	< 0.02	< 0.02	< 0.01	< 0.01	< 0.01
CH ₄	µmol/mol	2	0.031 ± 0.006	< 0.01	< 0.01	< 0.02	< 0.02	< 0.01	< 0.01	< 0.01
Non methane hydrocarbons	µmol/mol	2	< 0.05	< 0.02	< 0.02	< 0.02	< 0.02	0.156 ±	0.126 ±	0.111 ± 0.024
H ₂ O	µmol/mol	5	< 0.6	< 0.8	< 1.4	< 3	< 3	< 0.8	< 1.2	< 3
Total sulphur compounds	µmol/mol	0.004	< 0.002	< 0.002	< 0.002	< 0.002	< 0.002	< 0.0030	< 0.0030	< 0.0030
O ₂		5	0.45 ± 0.13	< 0.5	< 0.5	< 0.6	< 0.6	1.39 ± 0.36	< 0.5	1.59 ± 0.45
N ₂	µmol/mol	100	2.0 ± 0.5	4.6 ± 0.3	4.2 ± 0.4	< 1.5	< 1.5	1.51 ± 0.2	< 1.0	1.86 ± 0.2
Ar	µmol/mol	100	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5
Total halogenated (HCl)	µmol/mol	0.05	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	n.a.	< 0.005	< 0.005
CH ₂ O	µmol/mol	0.01	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O ₂	µmol/mol	0.2	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1
NH ₃	µmol/mol	0.1	< 0.1	< 0.1	< 0.1	< 0.1	< 0.1	n.a.	n.a.	n.a.
He	µmol/mol	300	34 ± 5	< 5	< 5	15 - 45	< 5	< 9	< 9	< 9
C2 hydrocarbons	µmol/mol	2	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	-	-	-
C3-hydrocarbons	µmol/mol	2	<1	<1	<1	<1	<1	-	-	-
C4-hydrocarbons	µmol/mol	2	<1	<1	<1	<1	<1	-	-	-
C5-hydrocarbons	µmol/mol	2	<1	<1	<1	<1	<1	-	-	-
C6 – C18 hydrocarbons	µmol/mol	2	<0.050	<0.05	<0.05	<0.05	<0.05	-	-	-

Hydrogen fuel quality from production methods - summary

- No contaminants observed above ISO 14687-2 threshold
 - 5 different sample from steam methane reformers with PSA
 - 8 different samples from PEM water electrolyser with TSA

- Challenges and additional contamination sources
 - Sampling
 - Hydrogen refuelling station infrastructure
 - Maintenance procedure

Sampling challenges

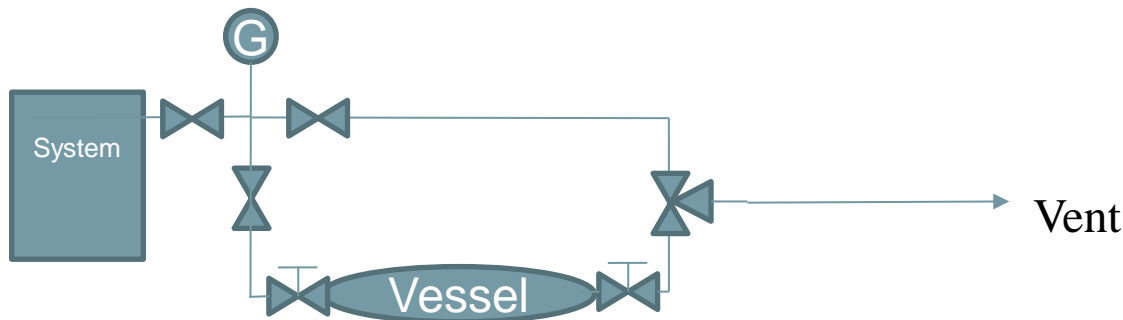
- Development of sampling strategy

- Sampling

- Pressure (14 – 100 bar)
 - Flow
 - Location (as close as production)
 - Sampling material
 - Sampling procedure



EMPIR Hydrogen consortium
Hydrogen producer procedure



- Cycling purge
- Purge through cylinder
- Cleanliness of cylinder
- Absence of leak

Example of contaminants observed in real samples

		Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)
CO	μmol/mol	< 0.02	< 0.02	< 0.053	< 0.02
CO ₂	μmol/mol	4.2 ± 0.3	< 0.01	0.101 ± 0.004	0.316 ± 0.007
CH ₄	μmol/mol	< 0.02	< 0.01	< 0.02	14.28 ± 0.07
Non methane hydrocarbons	μmol/mol	0.08 ± 0.01	0.111 ± 0.024	< 0.05	> 200
H ₂ O	μmol/mol	> 500	> 250	2.48 ± 0.25	13.2 ± 1.7
Total sulphur compounds	μmol/mol	< 0.002	< 0.0030	< 0.002	< 0.0036
O ₂	μmol/mol	> 520	1.59 ± 0.45	35 ± 2	< 0.5
N ₂	μmol/mol	3.7 ± 0.8	1.86 ± 0.2	134 ± 2	579 ± 23
Ar	μmol/mol	< 0.5	< 0.5	1.43 ± 0.10	< 1.0
Total halogenated (HCl)	μmol/mol	< 0.005	< 0.005	< 0.005	< 0.0036
CH ₂ O	μmol/mol	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O ₂	μmol/mol	< 0.1	< 0.1	< 0.1	< 0.1
NH ₃	μmol/mol	< 0.1	n.a.	< 0.1	< 0.1
He	μmol/mol	< 5	< 9	< 5	< 20

Electrolyser
without
purification

Sampling
contamination
(air, humidity)

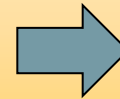
HRS maintenance
contamination
suspected

- No contaminants observed above ISO 14687-2 threshold from current production methods

- SMR with PSA samples
- PEM water electrolyser with TSA samples

- Challenges on contamination sources

- Sampling
- Hydrogen refuelling station infrastructure
- Maintenance procedure



Analytical evidences needed

- Report on risk assessment of impurities in hydrogen for fuel cells and recommendations on maximum concentration of individual compounds based on the new fuel cell degradation studies and on the probability of presence

Hydrogen

Hydrogen, as an energy source, is a clean and storable solution that could meet the worldwide energy demands.



Follow us at <https://projects.lne.eu/jrp-hydrogen/>

The new European policy objectives in the transport sectors defined in the Horizon 2020 Research and Innovation programme encourage the development of the hydrogen sector in order to reduce the greenhouse gas emissions.

The overall objective of the project is to address the standardisation needs in the hydrogen sector. The project is a direct result of the European Directive on the standardisation of hydrogen refuelling stations (2014/94/EU) in order to bring forward the metrology.

The project is contributing to the elaboration of two new standards that will supplement the revision of two ISO standards that will enable a sustainable implementation in the fast emerging hydrogen sector.

Revisions of these two ISO standards (ISO 11718:2015 Hydrogen – Part 2: Proton exchange membrane fuel cell systems for vehicles and ISO 16111:2008 Developing transport technologies – Hydrogen absorbed in reversible metal hydrides) are being developed by ISO/JTC 1 Hydrogen technologies and CEN/TC 268 Hydrogen technologies applications.

The two new standards are being developed in collaboration within the ISO/TC 197 standardization activities are ISO 21087 Hydrogen – Analytical methods – Proton exchange membrane (PEM) fuel cell applications for road vehicles and ISO 19880-8 Gaseous hydrogen – Fueling stations – Part 8: Fuel quality control.

The EMPIR project *Hydrogen* runs from 1 June 2016 to 31 May 2019.

A workshop related to the project is planned in November 2018. More info [here!](#)

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The EMPIR initiative is co-funded by the European Union's Horizon 2020 research and innovation programme and the EMPIR Participating States

NEWS

Workshop at Air Liquide R&D Centre: November 7 & 8, 2018

Hydrogen quality: publication in International Journal of Hydrogen Energy, April 2018

Upcoming events

Past events

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The EMPIR initiative is co-funded by the European Union's Horizon 2020 research and innovation programme and the EMPIR Participating States



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-  [Publication](#) in International Journal of Hydrogen Energy, April 2018
-  Flyer Hydrogen JRP
-  EURAMET 3rd Publishable Summary (January 2018)
-  [Publication](#) in Measurement