

Metrology research for hydrogen standardisation: a cross-cutting approach

Training course

Hydrogen purity measurements according to ISO 14687-2 and risk assessment for fuel cells

Training course - 07 November 2018

Air Liquide R&D – Versailles - France

- Overview of hydrogen quality requirements
- Quality control plan
- Monitoring / Sampling

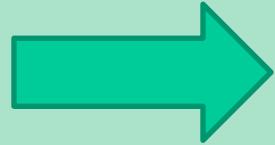
- Overview of hydrogen quality requirements – 5 min

- Quality control plan
 - Presentation – 15 min
 - Exercise – 20 min
 - Restitution – 10 min

- Monitoring / Sampling
 - Presentation – 10 min
 - Exercise – 10 min

- Summary / Discussion / Questions – 5 min

- Overview of hydrogen quality requirements
- Quality control plan
- Monitoring / Sampling



Compliance



Full scope according to standards
ISO 14687; SAE J2719:2011; EN 17124



Nozzle or agreed location



Quality control



Defined scope according to ISO 19880-8



Nozzle or agreed location



Analysis



Fit for purpose of analytical methods



Accuracy / uncertainty (ISO 21087)



Delay / price

Existing normative documents for fuel cell vehicles

Hydrogen Fuel Quality for Fuel Cell Vehicles

SAE J2719:2011



List of contaminants

Quality characteristics of hydrogen fuel for transport applications

ISO 14687-2: 2012



ISO/DIS 14687

List of contaminants
Nozzle sample

Hydrogen fuel - Product specification and quality assurance

EN 17124:2018



List of contaminants
Nozzle and other location
Risk assessment
"Level 1" value
Routine analysis

Fuel quality control

ISO/DIS 19880-8

Nozzle and other location
Risk assessment
"Level 1" value
Routine analysis

Gas analysis - Analytical methods for hydrogen fuel - Proton exchange membrane (PEM) fuel cell applications for road vehicles

ISO/DIS 21087

Analytical method fit for purpose
Method validation
Reporting

	ISO 14687: 2012 / SAE J2719:2011		ISO/CD 14687 / EN 17124	
	Max. admissible value [$\mu\text{mol/mol}$]	notes	Max. admissible value [$\mu\text{mol/mol}$]	notes
Water	5		5	
Total hydrocarbons (TC)	2	Due to CH_4 , TC > 2 $\mu\text{mol/mol}$	2 except CH_4	including oxygenated organic species
Methane	-		100	
Oxygen	5		5	
Helium	300		300	
Nitrogen	100	$\text{N}_2 + \text{Ar} < 100$	300	
Argon	100	$\text{N}_2 + \text{Ar} < 100$	300	
carbon dioxide	2		2	
Carbon monoxide	0.2		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Total sulphur compounds	0.004	H_2S , COS, CS_2 , mercaptans (NG)	0.004	H_2S , COS, CS_2 , mercaptans (NG)
Formaldehyde	0.01		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Formic acid	0.2		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Ammonia	0.1		0.1	
Halogenated compounds	0.05 (total)	i.e. HBr, HCl Cl_2 , organic R-X	0.05	HCl, organic R-Cl
Max. particulate conc.	1 mg/kg		1 mg/kg	

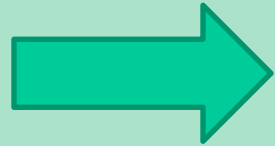
	ISO 14687: 2012 / SAE J2719:2011		ISO/CD 14687 / EN 17124	
	Max. admissible value [$\mu\text{mol/mol}$]	notes	Max. admissible value [$\mu\text{mol/mol}$]	notes
Water	5		5	
Total hydrocarbons (TC)	2	Due to CH_4 , TC > 2 $\mu\text{mol/mol}$	2 except CH_4	including oxygenated organic species
Methane	-		100	
Oxygen	5		5	
Helium	300		300	
Nitrogen	100	$\text{N}_2 + \text{Ar} < 100$	300	
Argon	100	$\text{N}_2 + \text{Ar} < 100$	300	
carbon dioxide	2		2	
Carbon monoxide	0.2		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Total sulphur compounds	0.004	H_2S , COS, CS_2 , mercaptans (NG)	0.004	H_2S , COS, CS_2 , mercaptans (NG)
Formaldehyde	0.01		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Formic acid	0.2		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Ammonia	0.1		0.1	
Halogenated compounds	0.05 (total)	i.e. HBr, HCl Cl_2 , organic R-X	0.05	HCl, organic R-Cl
Max. particulate conc.	1 mg/kg		1 mg/kg	

	ISO 14687: 2012 / SAE J2719:2011		ISO/CD 14687 / EN 17124	
	Max. admissible value [$\mu\text{mol/mol}$]	notes	Max. admissible value [$\mu\text{mol/mol}$]	notes
Water	5		5	
Total hydrocarbons (TC)	2	Due to CH_4 , TC > 2 $\mu\text{mol/mol}$	2 except CH_4	including oxygenated organic species
Methane	-		100	
Oxygen	5		5	
Helium	300		300	
Nitrogen	100	$\text{N}_2 + \text{Ar} < 100$	300	
Argon	100	$\text{N}_2 + \text{Ar} < 100$	300	
carbon dioxide	2		2	
Carbon monoxide	0.2		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Total sulphur compounds	0.004	H_2S , COS, CS_2 , mercaptans (NG)	0.004	H_2S , COS, CS_2 , mercaptans (NG)
Formaldehyde	0.01		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Formic acid	0.2		0.2	$\text{CO} + \text{HCHO} + \text{HCOOH} < 0.2 \mu\text{mol/mol}$
Ammonia	0.1		0.1	
Halogenated compounds	0.05 (total)	i.e. HBr, HCl Cl_2 , organic R-X	0.05	HCl, organic R-Cl
Max. particulate conc.	1 mg/kg		1 mg/kg	



Price: 2000 – 6000 € / sample
 Delay: 2 – 8 weeks

- Overview of hydrogen quality requirements
- Quality control plan
- Monitoring / Sampling



Compliance



Full scope according to standards
ISO 14687; SAE J2719:2011; EN17124



Regulatory requirements (Frequency
defined as part of ISO 19880-8)



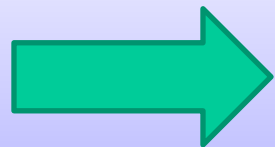
Quality control



prescriptive approach or Risk assessment



Frequency and list of key contaminants



Analysis



Price



Accuracy



Delay



Prescriptive approach methodology

Quality analysis of the contaminants listed in ISO 14687:

1 - Potential sources of contaminants

- o Sampling procedures
- o Characteristics of hydrogen production method(s)
- o Characteristics of hydrogen transport method(s)
- o Non-routine procedure (for example maintenance, major production system change)

2 - Analysis of possible contaminants

- o Possible quantification



Quality control defined

Prescriptive approach methodology

- **Production type**
- **Sampling point**
- **Frequency**
 - Reduced frequency
 - Online monitoring / operation control
 - Not satisfied → analysis once a day

Category: Distribution				
Facility type	Sampling/ Monitoring point	Contaminant	Threshold ($\mu\text{mol/mol}$)	Reduced frequency
Production of hydrogen from hydrocarbons utilizing steam reforming, catalytic reforming, partial oxidation, or ATR, purification using refining equipment, and distribution	Downstream of the purifier	TS ^a	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		N ₂ +Ar	100	Annual ^{b,c}
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^{b,c}
Electrolysis of Na Cl for hydrogen, purification, and distribution	Downstream of the purifier	Halogens	0,05	Annual ^b
		N ₂ +Ar	100	Annual ^b
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^b
Purification of coke-oven gas, and distribution	Downstream of the purifier	TS	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		Halogens	0,05	Annual ^b
		N ₂ +Ar	100	Annual ^b
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^b
		NH ₃	0,1	Annual ^b
		HCHO	0,01	Annual ^b
		HCOOH	0,2	Annual ^b
Purification of byproduct hydrogen from ethylene plants, and distribution	Downstream of the purifier	TS	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		N ₂ +Ar	100	Annual ^b
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^b

Prescriptive approach methodology

- Supply type
- Sampling point
- Frequency
 - Reduced frequency
 - Online monitoring / operation control
 - Not satisfied → analysis once a day

Category: Fueling station					
Facility type	Sampling/Monitoring point	Contaminant	Threshold (µmol/mol)	Reduced frequency	
With off-site supply of transported compressed or liquid hydrogen	End of nozzle	Those not analysed by the distributor		Annual ^b	
		N ₂ +Ar	100	Annual ^{b,c}	
		H ₂ O	5	Annual ^{b,d}	
		O ₂	5	Annual ^{b,e}	
With off-site supply from hydrogen pipelines	Downstream of the deodorant equipment	(Those listed for the odorant)		Annual ^{b,e}	
	End of nozzle	Those not analysed by the distributor		Annual ^b	
		N ₂ +Ar	100	Annual ^{b,c}	
		H ₂ O	5	Annual ^{b,d}	
With on-site supply of hydrogen produced from hydrocarbons utilizing steam reforming, catalytic reforming, partial oxidation, or ATR and purification using refining equipment	Downstream of the purifier	CO	0,2	Continuous and Annual ^{b,f}	
	End of nozzle	TS ^a	0,004	Annual ^b	
		THC as C1	2	Annual ^b	
		CO	0,2	Annual ^b	
		N ₂ +Ar	100	Annual ^{b,c}	
		H ₂ O	5	Annual ^{b,d}	
		O ₂	5	Annual ^{b,e}	
	With on-site supply from hydroelectrolysis and purification using refining equipment	Downstream of the purifier	N ₂ +Ar	100	Annual ^b
			H ₂ O	5	Continuous and Annual ^{b,f}
			O ₂	5	Continuous and Annual ^{b,f}
End of nozzle		Halogens	0,05	Annual ^b	
	N ₂ +Ar	100	Annual ^b		
	H ₂ O	5	Annual ^b		
	O ₂	5	Annual ^b		

Risk assessment methodology



Severity class: impact on fuel cell electrical vehicles

Table 2: Definition of severity classes

SEVERITY CLASS	FCEV Performance impact or damage	Impact categories		
		Performance impact	hardware impact temporary	Hardware impact permanent
0	No impact	No	No	No
1	Minor impact temporary loss of power No impact on hardware Car still operates	Yes	No	No
2	Reversible damage Requires specific procedure, light maintenance. Car still operates.	Yes or No	Yes	No
3	Reversible damage Requires specific procedure and immediate maintenance. Gradual power loss that does not compromises	Yes	Yes	No
4	Irreversible damage Requires major repair (e.g. stack change). Power loss or Car Stop that compromises safety	Yes	Yes	Yes

Severity class: impact on fuel cell electrical vehicles

Impurity		Severity Class for 0 ppm \leq Concentration $<$ ISO Value	ISO 14687-2 Threshold Value ² [ppm]	Severity Class for ISO Value \leq Concentration $<$ Level 1 Value	Level 1 Value [ppm]	Severity Class for Level 1 Value \leq Concentration \leq 100%
Total non-H ₂ gases		0	300	1	300	4
Total Nitrogen and Argon	N ₂ , Ar	0	100	1 ³	300 ³	4
Oxygen	O ₂	0	5	No test data available	No test data available	Without test data for proposed level 1 value validation already SC4 if ISO Spec exceeded
Carbon dioxide	CO ₂	0	2	1	3	4
Carbon monoxide	CO	0-2	0,2	2-3 ⁴	1	4
Methane	CH ₄	0	100	1	300	4
Water	H ₂ O	0	5	4	NA	Already SC4 if ISO Spec exceeded
Total sulfur compounds	H ₂ S basis	0-4	0.004	4	NA	Already SC4 if ISO Spec exceeded
Ammonia	NH ₃	0	0.1	4	NA	Already SC4 if ISO Spec exceeded
Total hydrocarbons	CH ₄ basis	0-4	2	1-4 ⁴	NA	Already SC4 if ISO Spec exceeded
Formaldehyde	CH ₂ O	0	0.01	2-3 ⁴	1	4
Formic Acid	CH ₂ O ₂	0-2	0.2	2-3 ⁴	1	4
Total halogenated compounds		0-4	0.05	4	NA	Already SC4 if ISO Spec exceeded
Helium	He	0	300	1	300	4
Maximum particulates concentration (liquid and solid)		0-4	1 mg/kg	4	NA	Already SC4 if ISO Spec exceeded



Guideline from ISO 19880-8

Influence on the severity class is complex and long term

Probability of occurrence of contaminants

Occurrence class	Class name	Occurrence or frequency	Occurrence or frequency
0	Very unlikely (practically impossible)	Contaminant above threshold never been observed for this type of source in the industry	Never
1	Very rare	Heard in the industry for the type of source / supply chain considered	1 per 1000000 refueling
2	Rare	Has happened more than once per year in the industry	1 per 100000 refueling
3	Possible	Has happened repeatedly for this type of source at a specific location	1 per 10000 refueling
4	Frequent	Happens on a regular basis	Often

Risk acceptability matrix

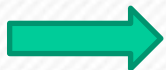
		Severity				
		0	1	2	3	4
Occurrence as the combined probabilities of the occurrence along the whole supply chain	4	Green	Red	Red	Red	Red
	3	Green	Yellow	Red	Red	Red
	2	Green	Green	Yellow	Red	Red
	1	Green	Green	Green	Yellow	Red
	0	Green	Green	Green	Green	Green
Key		Unacceptable risk: additional barriers are required	Further investigation needed: existing barriers or control may not be enough		Acceptable risk area: Existing control acceptable	



Guideline from ISO 19880-8

System defined

		Severity				
		0	1	2	3	4
Occurrence as the combined probabilities of the occurrence along the whole supply chain	4	Green	Red	Brown	Brown	Red
	3	Green	Yellow	Brown	Brown	Red
	2	Green	Green	Green	Brown	Red
	1	Green	Green	Green	Yellow	Red
	0	Green	Add barrier / Control		Green	Green
Key		Unacceptable risk: additional barriers are required	Further investigation needed: existing barriers or control may not be enough		Acceptable risk area: Existing control acceptable	

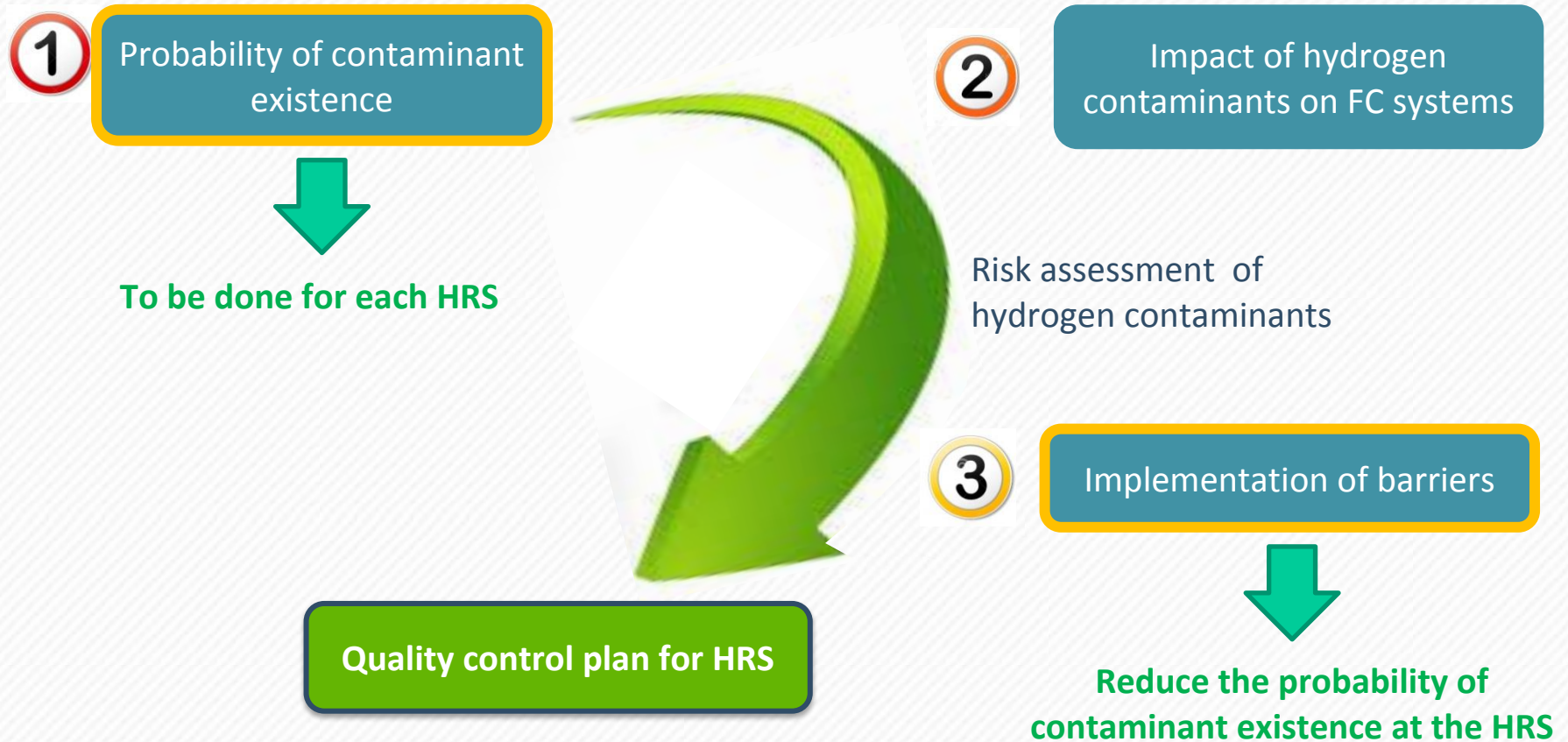


Each HRS will have different controls or barriers:

- Online sensors (i.e. hygrometers, CO sensor)
- Purification system
- Procedure (i.e. purging procedure after maintenance)
- Trained staff
-



Each HRS will have a different risk assessment

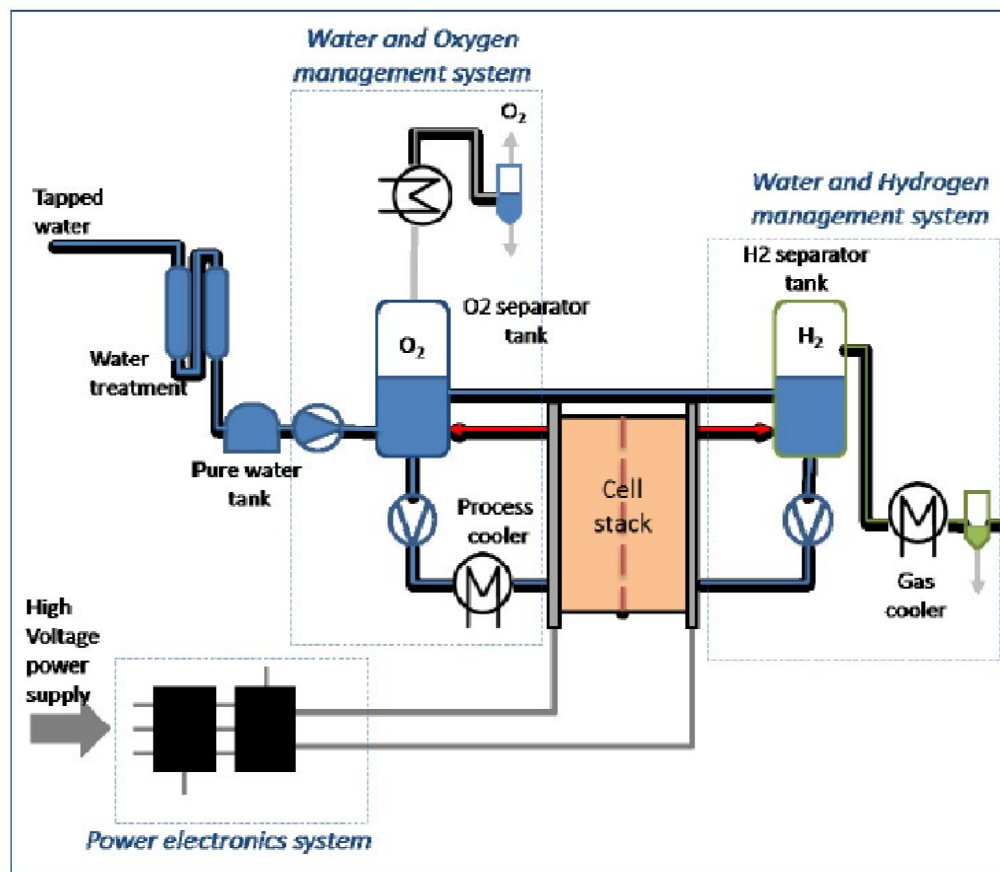


- Probability of contaminants presence on PEM water electrolyser
- Implementation of barriers
- Monitoring

Example / Exercise

PEM Water electrolyser

Compounds	ISO 14687-2 threshold [μmol/mol]	PEM water electrolysis (analysis) Results [μmol/mol]	Probability of occurrence [0-4]
Water H ₂ O	5	> 100	
Methane CH ₄	2	< 0.02 – 0.1	
Non CH ₄ hydrocarbons	2	< 0.02 – 0.09	
Oxygen O ₂	5	18- > 500	
Helium He	300	< 9	
Nitrogen N ₂	100	1.2 – 4.5	
Argon Ar	100	< 0.5	
Carbon dioxide CO ₂	2	0.2 – 5.4	
Carbon monoxide CO	0.2	< 0.02	
Total sulphur compounds	0.004	< 0.0036	
Formaldehyde HCHO	0.01	< 0.005	
Formic acid HCOOH	0.2	< 0.1	
Ammonia NH ₃	0.1	< 0.1	
Total halogenated	0.05	< 0.005	

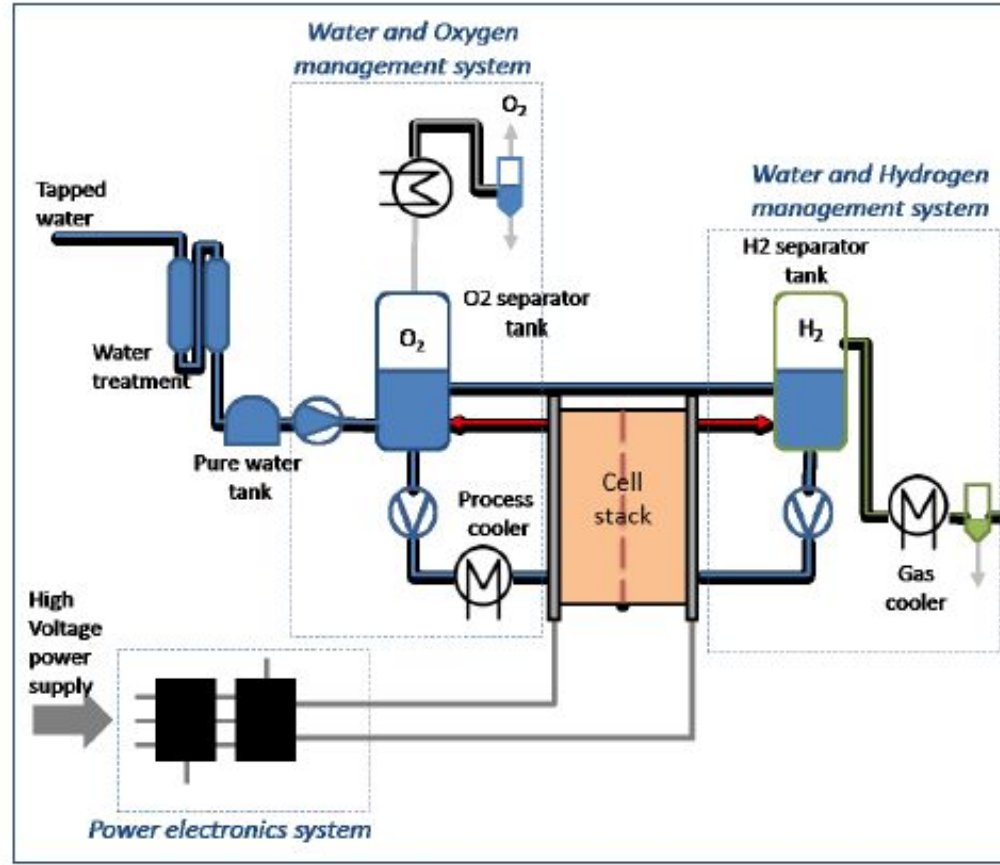


Question 1: What is the probability of contaminants presence on PEM water electrolyser (reply directly in the table)?

Example / Exercise

PEM Water electrolyser

Compounds	ISO 14687-2 threshold [$\mu\text{mol/mol}$]	PEM water electrolysis (analysis) Results [$\mu\text{mol/mol}$]	Probability of occurrence [0-4]
Water H_2O	5	> 100	
Methane CH_4	2	< 0.02 – 0.1	
Non CH_4 hydrocarbons	2	< 0.02 – 0.09	
Oxygen O_2	5	18- > 500	
Helium He	300	< 9	
Nitrogen N_2	100	1.2 – 4.5	
Argon Ar	100	< 0.5	
Carbon dioxide CO_2	2	0.2 – 5.4	
Carbon monoxide CO	0.2	< 0.02	
Total sulphur compounds	0.004	< 0.0036	
Formaldehyde HCHO	0.01	< 0.005	
Formic acid HCOOH	0.2	< 0.1	
Ammonia NH_3	0.1	< 0.1	
Total halogenated	0.05	< 0.005	



Question 2: What controls and barriers can be implemented based on your probability of occurrence?

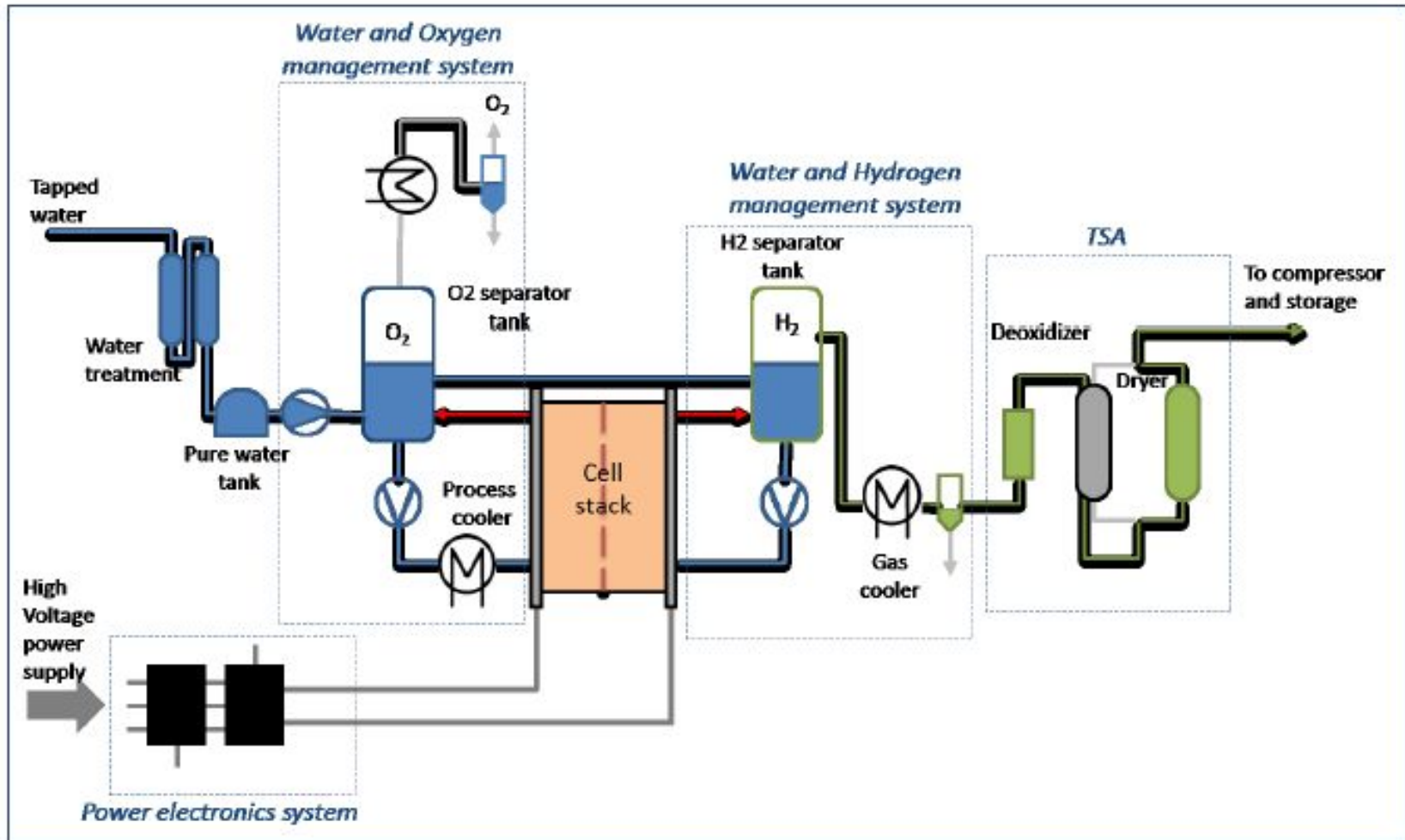
Example / Exercise

PEM Water electrolyser

Compounds	ISO 14687-2 threshold [μmol/mol]	PEM water electrolysis (analysis) Results [μmol/mol]	Probability of occurrence [0-4]	Controls and barriers	Probability of occurrence after control and barrier [0-4]
Water H ₂ O	5	> 100			
Methane CH ₄	2	< 0.02 – 0.1			
Non CH ₄ hydrocarbons	2	< 0.02 – 0 .09			
Oxygen O ₂	5	18- > 500			
Helium He	300	< 9			
Nitrogen N ₂	100	1.2 – 4.5			
Argon Ar	100	< 0.5			
Carbon dioxide CO ₂	2	0.2 – 5.4			
Carbon monoxide CO	0.2	< 0.02			
Total sulphur compounds	0.004	< 0.0036			
Formaldehyde HCHO	0.01	< 0.005			
Formic acid HCOOH	0.2	< 0.1			
Ammonia NH ₃	0.1	< 0.1			
Total halogenated	0.05	< 0.005			

Example / Exercise

PEM Water electrolyser



Question 3: What controls and barriers would you add on this system?

Reply directly on the schematic

Question 4: What differences do you observe between your probability of occurrence and this results?

Probability of impurity presence	PEM water electrolysis process with TSA
Frequent	None identified
Possible	None identified
Rare	N ₂ , O ₂ , H ₂ O
Very Rare	CO ₂
Unlikely	He, Ar, CO, CH ₄ , HCHO, HCOOH, NH ₃ , sulfur compounds, hydrocarbons (except CH ₄), halogenated compounds

Differences:

Explanations:

Discussion / Feedback from groups

- Overview of hydrogen quality requirements
- Quality control plan
- Monitoring / Sampling

Hydrogen quality monitoring

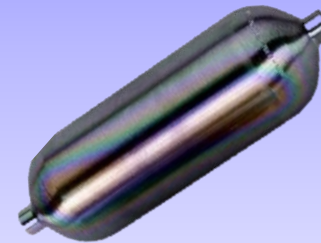
- **Online analysers / Sensors**

- Pressure
- Flow
- Location
- Maintenance
- Calibration



- **Sampling**

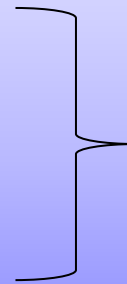
- Pressure
- Flow
- Location
- Sampling material
- Sampling procedure



Hydrogen quality monitoring

- **Online analysers / Sensors**

- Pressure
- Flow
- Location
- Maintenance
- Calibration



Humidity

O₂

CO

CH₄

CO₂

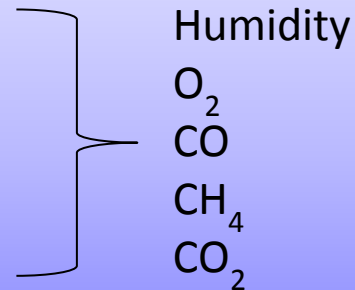
- **Sampling**

- Pressure
- Flow
- Location
- Sampling material
- Sampling procedure

Hydrogen quality monitoring

- **Online analysers / Sensors**

- Pressure
- Flow
- Location
- Maintenance
- Calibration



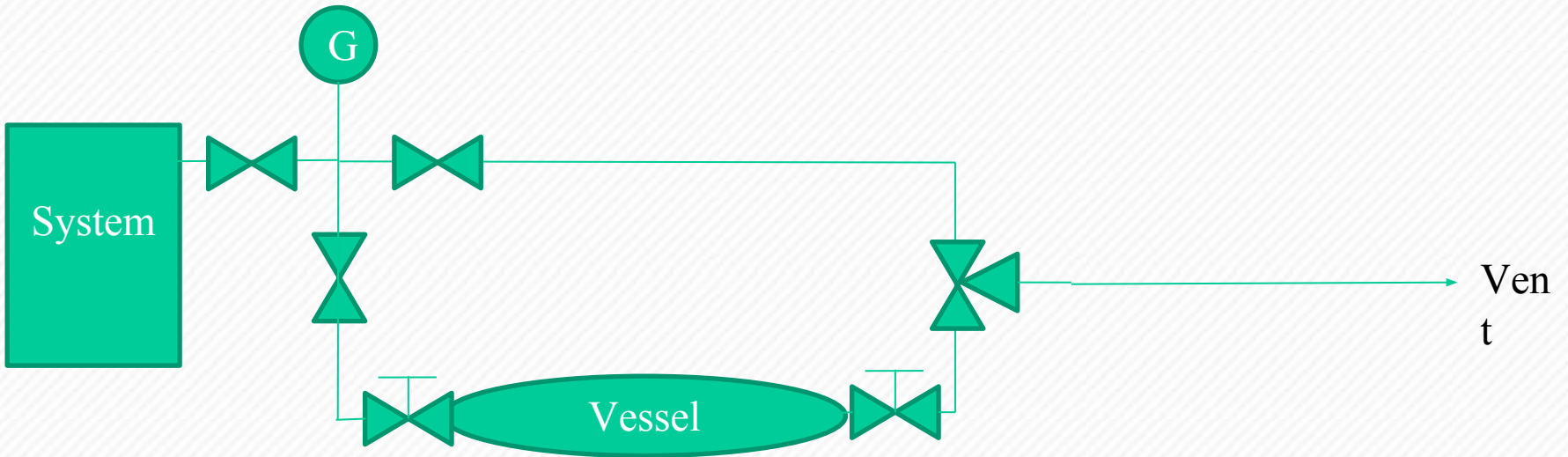
Humidity
O₂
CO
CH₄
CO₂

- **Sampling**

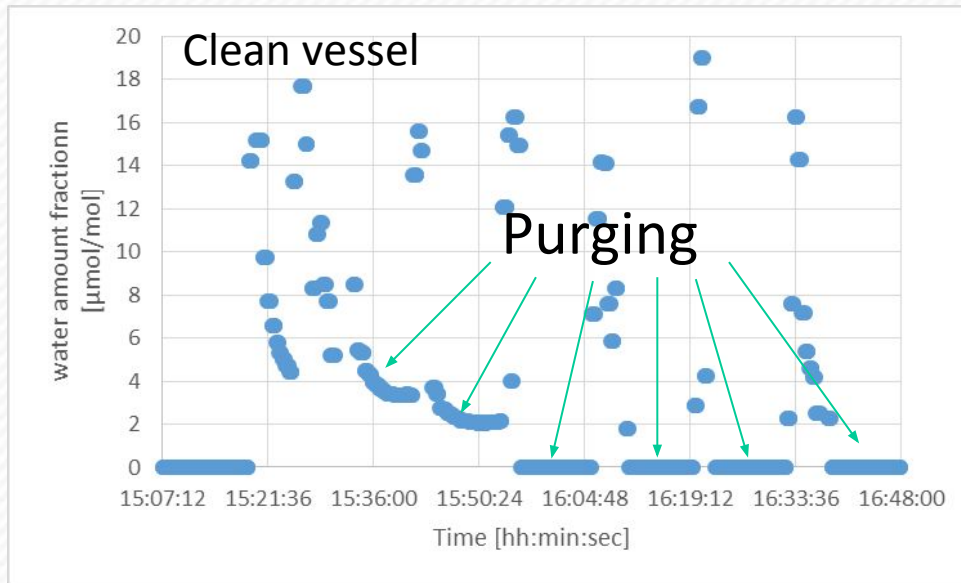
- Pressure
- Flow
- Location
- Sampling material
- Sampling procedure



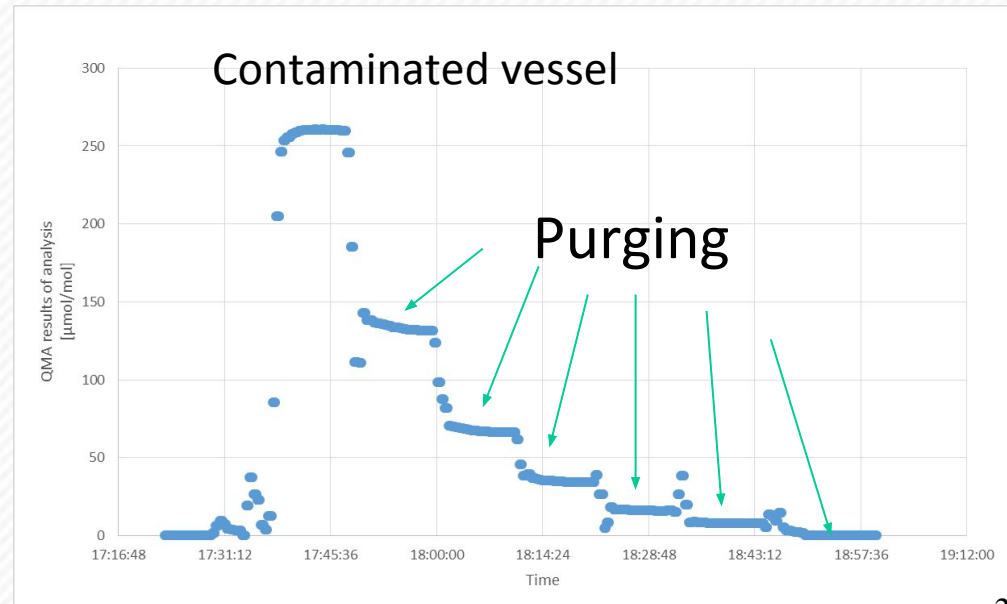
**Eliminating
contamination**



Monitoring / Sampling Humidity – water vapour



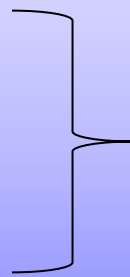
- Cycling purge
- Purge through cylinder
- Cleanliness of cylinder
- Absence of leak



Hydrogen quality monitoring

- **Online analysers / Sensors**

- Pressure
- Flow
- Location
- Maintenance
- Calibration



Humidity
O₂
CO
CH₄
CO₂

- **Sampling**

- Pressure
- Flow
- Location
- Sampling material
- Sampling procedure
- Material of sampling system (passivated)
- Leak check
- Purge
- Procedure
- Trained staff
- Cylinder (transportable, passivated)



Example of issues with analytical results

		Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)
CO	μmol/mol	< 0.02	< 0.02	< 0.053	< 0.053
CO ₂	μmol/mol	4.2 ± 0.3	< 0.01	0.101 ± 0.004	0.443 ± 0.010
CH ₄	μmol/mol	< 0.02	< 0.01	< 0.02	0.031 ± 0.006
Non methane hydrocarbons	μmol/mol	0.08 ± 0.01	0.111 ± 0.024	< 0.05	< 0.05
H ₂ O	μmol/mol	> 500	> 250	2.48 ± 0.25	< 0.6
Total sulphur compounds	μmol/mol	< 0.002	< 0.0030	< 0.002	< 0.002
O ₂	μmol/mol	> 520	1.59 ± 0.45	35 ± 2	0.45 ± 0.13
N ₂	μmol/mol	3.7 ± 0.8	1.86 ± 0.2	134 ± 2	2.0 ± 0.5
Ar	μmol/mol	< 0.5	< 0.5	1.43 ± 0.10	< 0.5
Total halogenated (HCl)	μmol/mol	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O	μmol/mol	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O ₂	μmol/mol	< 0.1	< 0.1	< 0.1	< 0.1
NH ₃	μmol/mol	< 0.1	n.a.	< 0.1	< 0.1
He	μmol/mol	< 5	< 9	< 5	< 5

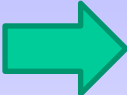
Question 1: What potential issues do you find in these analysis?
Highlight the results of analysis that are suspicious

Question 2: Is there results of analysis that were contaminated by the sampling process?

Example of issues with analytical results


		ISO 14687-2 threshold	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)	Results with expanded uncertainty (k=2)
CO	μmol/mol	0.2	< 0.02	< 0.02	< 0.053	< 0.053
CO ₂	μmol/mol	2	4.2 ± 0.3	< 0.01	0.101 ± 0.004	0.443 ± 0.010
CH ₄	μmol/mol	2	< 0.02	< 0.01	< 0.02	0.031 ± 0.006
Non methane hydrocarbons	μmol/mol	2	0.08 ± 0.01	0.111 ± 0.024	< 0.05	< 0.05
H ₂ O	μmol/mol	5	> 500	> 250	2.48 ± 0.25	< 0.6
Total sulphur compounds	μmol/mol	0.004	< 0.002	< 0.0030	< 0.002	< 0.002
O ₂	μmol/mol	5	> 520	1.59 ± 0.45	35 ± 2	0.45 ± 0.13
N ₂	μmol/mol	100	3.7 ± 0.8	1.86 ± 0.2	134 ± 2	2.0 ± 0.5
Ar	μmol/mol	100	< 0.5	< 0.5	1.43 ± 0.10	< 0.5
Total halogenated (HCl)	μmol/mol	0.05	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O	μmol/mol	0.01	< 0.005	< 0.005	< 0.005	< 0.005
CH ₂ O ₂	μmol/mol	0.2	< 0.1	< 0.1	< 0.1	< 0.1
NH ₃	μmol/mol	0.1	< 0.1	n.a.	< 0.1	< 0.1
He	μmol/mol	300	< 5	< 9	< 5	< 5
Reply question 2			Normal results or contaminated during sampling	Normal results or contaminated during sampling	Normal results or contaminated during sampling	Normal results or contaminated during sampling
Reasons						

Hydrogen monitoring

- Sampling
 - Pressure
 - Flow
 - Location
 - Sampling material
 - Sampling procedure
 - Material of sampling system (passivated)
 - Leak check
 - Purge
 - Procedure
 - Trained staff
 - Cylinder (transportable, passivated)
- 



Typical issues:

- Air (ratio N_2 / O_2 / Ar)
 - Humidity (high content)
- 
- Leak check
 - Purge
 - Procedure
 - Trained staff

- **Overview of hydrogen quality requirements**

- ISO 14687 Hydrogen quality
- ISO 19880-8 Hydrogen quality control

- **Quality control plan**

- Prescriptive
- Risk assessment → for each HRS
 - Severity (fixed)
 - Probability of contaminant presence
 - Depend on system and barriers

- **Monitoring / Sampling**

- Online
- Sampling
 - Sampling system (passivated); cylinder (transportable, passivated)
 - Leak check
 - Purge
 - Procedure and trained staff

Hydrogen

Hydrogen

Hydrogen, as an energy source, is a clean and storable solution that could meet the worldwide energy demands.



Follow us at <https://projects.lne.eu/jrp-hydrogen/>
 frederique.haloua@lne.fr

The new European policy objectives in the transport sectors defined in the Horizon 2020 Research and Innovation programme encourage the development of the hydrogen sector in order to reduce the greenhouse gas emissions.

The overall objective of the project is to address the standardisation needs in the hydrogen sector in line with the requirements of the European Directive on the standardisation of hydrogen refuelling stations (2014/94/EU) in order to bring forward the metrology.

The project aims at supplementing the revision of two ISO standards that are currently under development to enable a sustainable implementation in the fast emerging hydrogen market contributing to the elaboration of two new standards.

Revisions of these two ISO standards (ISO 14303:2013 *Hydrogen technologies – Part 2: Proton exchange membrane fuel cell (PEMFC) systems for vehicles* and ISO 16111:2008 *Developing transport technologies – Hydrogen absorbed in reversible metal hydrides*) are currently under development under the ISO/TC 197 *Hydrogen technologies and applications* and CEN/TC 268 *Hydrogen technologies applications*.

The two new standards are currently under development within the ISO/TC 197 standardization activities are ISO 21087 *Hydrogen technologies – Analytical methods – Proton exchange membrane (PEM) fuel cell applications for road vehicles* and ISO 19880-8 *Gaseous hydrogen – Fueling stations – Part 8: Fuel quality control*.

The EMPIR project *Hydrogen* runs from 1 June 2016 to 31 May 2019.

A workshop related to the project is planned in November 2018. More info [here!](#)



NEWS

Workshop at Air Liquide R&D Centre: November 7 & 8, 2018

Hydrogen quality: publication in International Journal of Hydrogen Energy, April 2018

Upcoming events

Past events

DOWNLOAD

- [Publication](#) in International Journal of Hydrogen Energy, April 2018
- Flyer Hydrogen JRP
- EURAMET 3rd Publishable Summary (January 2018)
- [Publication](#) in Measurement



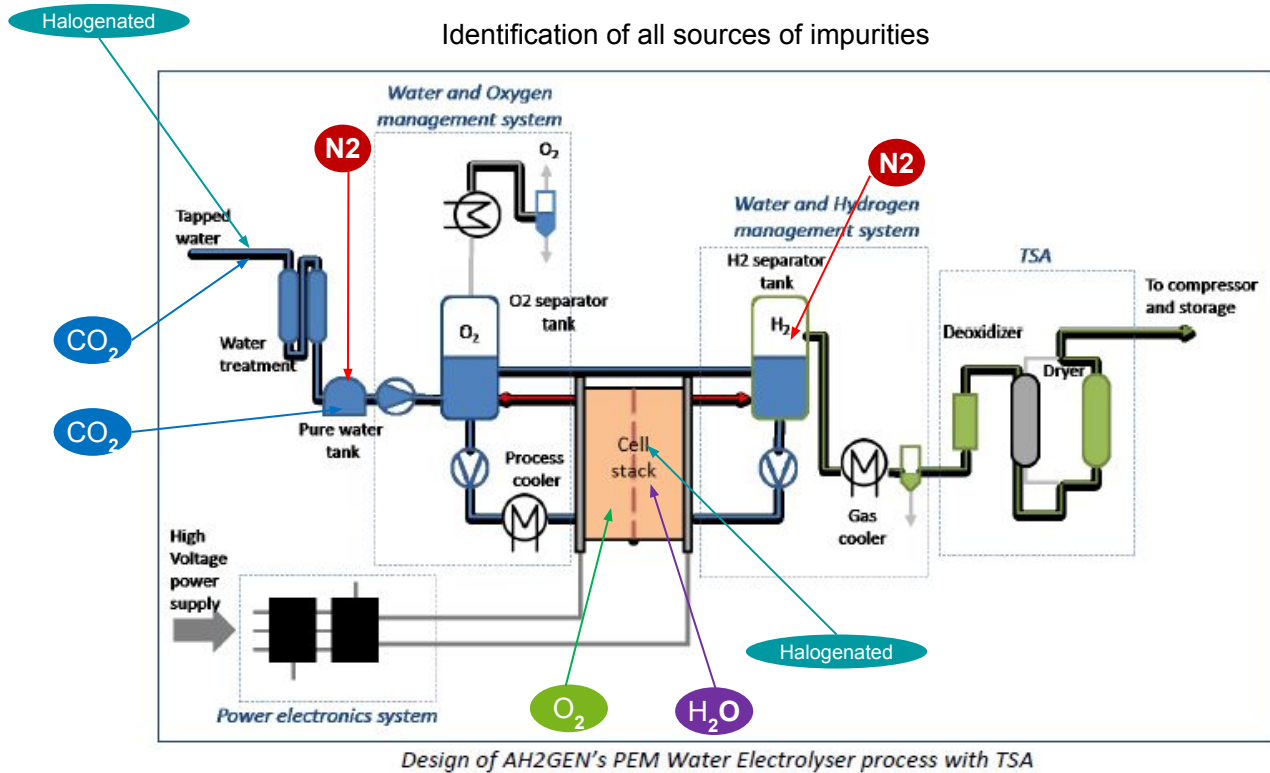
THE PROJECT



WORKPACKAGES



2 – PEM Electrolysis + TSA



2 – PEM Electrolysis + TSA

Identification of all existing barriers

